# ETHYLENE GLYCOL

STABILITY AND REACTIVITY

Chemical Stability

Stable at normal temperatures and storage conditions.

Incompatibility with Other Materials

None reasonably foreseeable.

Decomposition

Decomposition will not occur.

Polymerization

Polymerization can occur. (Not violent or strongly exothermic.) Extended heating at high temperatures (>200 degC).

TOXICOLOGICAL INFORMATION

Animal Data

1,2-Ethanediol:

Oral LD50: 4,000 mg/kg in female rats Dermal LD50: >20 mL/kg in rabbits

SIPEG did not produce genetic damage in bacterial cell cultures.

1,2-Ethanediol is a mild skin irritant and mild eye irritant, and is untested for skin sensitization in animals. Repeated exposure by ingestion caused histopathological changes of the kidneys, bone marrow, kidney effects with oxalate crystal deposition, altered hematology, decreased body weight. Long-term exposure caused kidney effects with oxalate crystal deposition, histopathological changes of the kidneys, liver, blood vessels, testes, sperm, decreased body weight. No deaths occurred in animals exposed to saturated vapors of the compound. Repeated exposure by inhalation caused histopathological changes of the liver, lungs, eye

# ETHYLENE GLYCOL

irritation, clouding of the eye (corneal opacity). In animal testing this material has not caused carcinogenicity. Reproductive data on adult animals show interference with reproduction only at levels which produce other toxic effects in the adult animal. Tests have shown this material to cause developmental toxicity in animals. This material has not produced genetic damage in bacterial cultures. There are reports indicating that this material does not produce genetic damage in some animal or mammalian cell culture tests; however, there are reports in the literature that suggest positive results.

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# ECOLOGICAL INFORMATION

# Ecotoxicological Information

Ethylene Glycol:

96 hour LC50, Fathead minnows: 49,000 mg/L

# DISPOSAL CONSIDERATIONS

# Waste Disposal

Treatment, storage, transportation, and disposal must be in accordance with applicable Federal, State/Provincial, and Local regulations.

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# TRANSPORTATION INFORMATION

# Shipping Information

DOT/IMO/IATA : Not Regulated in Containers with

less than 5,000 lbs. Ethylene Glycol

If greater than 5,000 lbs. Ethylene Glycol, use:

DOT/IMO/IATA

Proper Shipping Name : Environmentally Hazardous Substance,

Liquid, N.O.S.

(Contains Ethylene Glycol)

\_\_\_\_\_

Hazard Class : 9

# ETHYLENE GLYCOL

UN Number : 3082
Packing Group : III
Label : Class 9

Reportable Quantity : 5,000 lbs. Ethylene Glycol

# Shipping Information -- Canada

This material is Not Regulated.

# REGULATORY INFORMATION

# U.S. Federal Regulations

TSCA Inventory Status : Listed.

TITLE III HAZARD CLASSIFICATIONS SECTIONS 311, 312

Acute : Yes Chronic : Yes Fire : No Reactivity : No Pressure : No

# Canadian Regulations

This product has been classified in accordance with the hazard criteria of the CPR and the MSDS contains all the information required by the CPR.

Not listed on the Canadian Domestic Substances List (DSL).

# OTHER INFORMATION

# NFPA, NPCA-HMIS

NPCA-HMIS Rating

Health : 2
Flammability : 1
Reactivity : 0

Personal Protection rating to be supplied by user depending on use conditions.

# Additional Information

This product contains polymer-grade ethylene glycol.

The data in this Material Safety Data Sheet relates only to the

The data in this Material Safety Data Sheet relates only to the specific material designated herein and does not relate to use in combination with any other material or in any process.

# ETHYLENE GLYCOL

Responsibility for MSDS

CHEMICALS
DuPont Canada Inc.
7070 Mississauga Rd.
Mississauga, Ontario, L5M 2H3
(905) 821-5369.

End of MSDS

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# FREON 502

\* Canadian Centre for Occupational Health and Safety \*
\* \* \* \* \* \* \* \* \* \* \* Issue : 2001-1 (February, 2001) \*

\*\*\* IDENTIFICATION \*\*\*

MSDS RECORD NUMBER

PRODUCT NAME(S) : CHLORODIFLUOROMETHANE &

CHLOROPENTAFLUOROETHANE MIXTURE

"FREON" 502

PRODUCT IDENTIFICATION : MSDS NUMBER: CEF00502

DATE OF MSDS : 1999-05-20

CURRENCY NOTE : This MSDS was provided to CCOHS in

: 2438327

electronic form on 2000-10-30

\*\*\* MANUFACTURER INFORMATION \*\*\*

MANUFACTURER : DuPont Canada, Inc

ADDRESS : Post Office Box 2200

Streetsville

Mississauga Ontario Canada L5M 2H3

Telephone: 800-387-2122 (Product

Information)

EMERGENCY TELEPHONE NO.: 613-348-3616 (Transport, 24 HOURS)

613-348-3616 (Medical, 24 HOURS)

\*\*\* SUPPLIER/DISTRIBUTOR INFORMATION \*\*\*

SUPPLIER/DISTRIBUTOR : DuPont Canada, Inc

ADDRESS : F

: Post Office Box 2200

Streetsville

Mississauga Ontario

Canada L5M 2H3

Telephone: 800-387-2122 (Product

Information)

EMERGENCY TELEPHONE NO.: 613-348-3616 (Transport, 24 HOURS)

613-348-3616 (Medical, 24 HOURS)

FREON 502

CHEMICAL PRODUCT/COMPANY IDENTIFICATION

Material Identification

"FREON" is a registered trademark of DuPont.

Corporate MSDS Number : DU001047

Product Use

Refrigerant

Tradenames and Synonyms

CHLORODIFLUOROMETHANE & CHLOROPENTAFLUOROETHANE MIXTURE

(AZEOTROPE)

Company Identification

MANUFACTURER/DISTRIBUTOR

DuPont Canada, Inc. P.O. Box 2200

Streetsville

Formula : CHClF2/CClF2CF3

Mississauga, Ontario L5M 2H3

PHONE NUMBERS

Product Information : 1-800-387-2122

Transport Emergency : 1-613-348-3616 (24 HOURS)
Medical Emergency : 1-613-348-3616 (24 HOURS)

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# COMPOSITION/INFORMATION ON INGREDIENTS

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Components

Material CAS Number %
\*ETHANE, CHLOROPENTAFLUORO 76-15-3 51.2 WT%
("FREON" 115)
\*METHANE, CHLORODIFLUORO 75-45-6 48.8 WT%
("FREON" 22)

<sup>\*</sup> Disclosure as a toxic chemical is required under Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR part 372.

		=======================================	= :
	FREON	v 502	
			==
HAZARDS IDENTIF	CATION		

# Potential Health Effects

Inhalation of high concentrations of vapor is harmful and may cause heart irregularities, unconsciousness, or death. Intentional misuse or deliberate inhalation can cause death without warning. Vapor reduces oxygen available for breathing and is heavier than air. Liquid contact can cause frostbite.

# HUMAN HEALTH EFFECTS:

Overexposure to the vapors by inhalation may include temporary nervous system depression with anesthetic effects such as dizziness, headache, confusion, incoordination, and loss of consciousness. Higher inhalation overexposures to the vapors may cause temporary alteration of the heart's electrical activity with irregular pulse, palpitations, or inadequate circulation. Fatality from gross overexposure may occur. Skin contact with the liquid may cause frostbite.

Individuals with preexisting diseases of the central nervous system, cardiovascular system, lungs or kidneys may have increased susceptibility to the toxicity of excessive exposures.

# Carcinogenicity Information

None of the components present in this material at concentrations equal to or greater than 0.1% are listed by IARC, NTP, OSHA or ACGIH as a carcinogen.

# FIRST AID MEASURES

First Aid

# INHALATION

If large amounts are inhaled, immediately remove to fresh air. Keep persons calm. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.

# SKIN CONTACT

In case of skin contact, flush with water for 15 minutes. Treat for frostbite if necessary by gently warming affected area.

# FREON 502

# EYE CONTACT

In case of eye contact, immediately flush eyes with plenty of water for 15 minutes. Call a physician.

# INGESTION

Ingestion is not considered a potential route of exposure.

# Notes to Physicians

Because of possible disturbances of cardiac rhythm, catecholamine drugs, such as epinephrine, should be used with special caution only in situations of emergency life support.

# FIRE FIGHTING MEASURES

Flammable Properties

: Will not burn Flash Point Flammable limits in Air, % by Volume : Not applicable LEL UEL : Not applicable Autoignition : 704 C (1299 F)

Fire and Explosion Hazards:

Cylinders are equipped with temperature and pressure relief devices but still may rupture under fire conditions. Decomposition may occur.

Extinguishing Media

As appropriate for combustibles in area.

Fire Fighting Instructions

Keep containers cool with water spray. Self-contained breathing apparatus (SCBA) is required if cylinders rupture or release under fire conditions.

# FREON 502 ACCIDENTAL RELEASE MEASURES Safeguards (Personnel) NOTE: Review FIRE FIGHTING MEASURES and HANDLING (PERSONNEL) sections before proceeding with clean-up. Use appropriate PERSONAL PROTECTIVE EQUIPMENT during clean-up. Accidental Release Measures Ventilate area - especially low places where heavy vapors might collect. Remove open flames. Use self-contained breathing apparatus (SCBA) for large spills. Comply with Federal, State, and local regulations for reporting

# HANDLING AND STORAGE

Handling (Personnel)

releases.

Avoid breathing vapors. Avoid liquid contact with skin or eyes. Use with sufficient ventilation to keep employee exposure below recommended limits.

Storage

Clean, dry area. Do not heat above 52 deg C (125 deg F).

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# EXPOSURE CONTROLS/PERSONAL PROTECTION

Engineering Controls

Use with sufficient ventilation to keep employee exposure below recommended exposure limits. Local exhaust should be used when large amounts are released. Mechanical ventilation should be used in low or enclosed places.

Personal Protective Equipment

Impervious gloves and chemical splash goggles should be used if contact is possible. Under normal manufacturing conditions, no respiratory protection is required when using this product. Self-contained breathing apparatus (SCBA) is required if a spill or release occurs.

# FREON 502

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# Exposure Guidelines

Applicable Exposure Limits ETHANE, CHLOROPENTAFLUORO

PEL (OSHA) : None Established
TLV (ACGIH) : 1,000 ppm, 6,320 mg/m3, 8 Hr. TWA
AEL \* (DuPont) : None Established

METHANE, CHLORODIFLUORO

: None Established

PEL (OSHA) TLV (ACGIH) PEL (OSHA) : None Established

TLV (ACGIH) : 1,000 ppm, 3,540 mg/m3, 8 Hr. TWA, A4

AEL \* (DuPont) : None Established

\* AEL is DuPont's Acceptable Exposure Limit. Where governmentally imposed occupational exposure limits which are lower than the AEL are in effect, such limits shall take precedence.

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# PHYSICAL AND CHEMICAL PROPERTIES

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Physical Data

Boiling Point : -45.4 C (-49.7 F)

Vapor Pressure : 169 psia at 25 deg C (77 deg F)

Vapor Density : 3.92 at 25 deg C (77 deg F) (Air= 1)

% Volatiles : 100 WT%

% Volatiles : 100 MIC
Evaporation Rate : >1 (CCl4 = 1)
Solubility in Water : 0.15 WT% @ 25 C (77 F)
: Neutral

Odor : Slight ethereal

: Liquified gas Form

: Clear, colorless Color

: 1.22 g/cc at 25 deg C (77 deg F) - Liquid Density

# 

# STABILITY AND REACTIVITY

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Chemical Stability

Material is stable. However, avoid open flames and high temperatures.

Incompatibility with Other Materials

Incompatible with alkali or alkaline earth metals- powdered

# FREON 502

Al, Zn, Be, etc.

Polymerization

Polymerization will not occur.

Other Hazards

Decomposition : Decomposition products are hazardous. "FREON" 502 Refrigerant can be decomposed by high temperatures (open flames, glowing metal surfaces, etc.) forming hydrochloric and hydrofluoric acids, and possibly carbonyl halides.

# TOXICOLOGICAL INFORMATION

Animal Data

"FREON" 115

Inhalation 4-hour LC50: >800,000 ppm in rats Oral ALD : >1200 mg/kg in rats

The effects of a single inhalation exposure at high concentrations include rapid respiration and inactivity. Repeated exposure at lower levels produced no signs of toxicity. Exposure to 150,000 ppm with simultaneous epinephrine challenge produced cardiac arrhythmia in dogs. The effects of repeated ingestion include mild diarrhea, salivation and increased activity.

No animal test reports are available to define carcinogenic developmental or reproductive hazards. The compound does not produce genetic damage in bacterial cell cultures but has not been tested in animals.

"FREON" 22

Inhalation 4-hour LC50: 220,000 ppm in rats

The compound is a skin irritant and a slight eye irritant, but is not a skin sensitizer in animals.

Effects from single high exposures include central nervous system depression, anesthesia, rapid breathing, lung congestion and microscopic liver changes. Cardiac sensitization occurred in dogs at 50,000 ppm or greater from the action of exogenous epinephrine.

# FREON 502

No toxic effects or abnormal histopathological observations occurred in rats repeatedly exposed to concentrations ranging from 10,000 to 50,000 ppm (v/v). Long-term exposures to 50,000 ppm (v/v) of vapors produced organ weight increases and a decrease in body weight gain, but no increased mortality or adverse hematological effects.

In chronic inhalation studies, HCFC-22, at a concentration of 50,000 ppm (v/v), produced a small, but statistically significant increase of late-occurring tumors involving salivary glands in male rats, but not female rats or male or female mice. In the same studies, no increased incidence of tumors was seen in either species at concentrations of 10,000 ppm or 1000 ppm (v/v).

Long-term administration in corn oil produced no effects on body weight or mortality.

HCFC-22 was mutagenic in some strains of bacteria in bacterial cell cultures, but not mammalian cell cultures or animals. It did not cause heritable genetic damage in mammals.

A slight, but significant increase in developmental toxicity was observed at high concentrations (50,000 ppm) of HCFC-22, a concentration which also produced toxic effects in the adult animal. Based on these findings, and other negative developmental studies, HCFC-22 is not considered a unique hazard to the conceptus. Studies of the effects of HCFC-22 on male reproductive performance have been negative. Specific studies to evaluate the effect on female reproductive performance have not been conducted, however, limited information obtained from studies on developmental toxicity do not indicate adverse effects on female reproductive performance at concentrations up to 50,000 ppm.

# ECOLOGICAL INFORMATION

Ecotoxicological Information

Aquatic Toxicity:

"Freon" 22

48 hour EC50 - Daphnia magna: 433 mg/L

# FREON 502

# DISPOSAL CONSIDERATIONS

Waste Disposal

Comply with Federal, State, and local regulations. Remove to a permitted waste disposal facility or reclaim by distillation.

# TRANSPORTATION INFORMATION

Shipping Information

DOT/IMO

Proper Shipping Name : CHLORODIFLUOROMETHANE AND

CHLOROPENTAFLUOROETHANE MIXTURE

Hazard Class 2.2 UN No. : 1973

: NONFLAMMABLE GAS DOT/IMO Label

\_\_\_\_\_\_

Shipping Containers

Cylinders Ton Tanks

Shipping Information -- Canada

TDG

Proper Shipping Name : CHLORODIFLUOROMETHANE and

CHLOROPENTAFLUOROETHANE Mixture

UN # : 1973 TDG Class : 2.2

# REGULATORY INFORMATION

U.S. Federal Regulations

TSCA Inventory Status : Reported/Included.

# FREON 502

TITLE III HAZARD CLASSIFICATIONS SECTIONS 311, 312

Acute : Yes Chronic : No Fire : No Reactivity : No Pressure : Yes

HAZARDOUS CHEMICAL LISTS

SARA Extremely Hazardous Substance: No CERCLA Hazardous Substance : No

SARA Toxic Chemical - See Components Section

Canadian Regulations

CEPA Status : DSL: REPORTED/INCLUDED.

WHMIS Classification:

CLASS A Compressed Gas

This product has been classified in accordance with the hazard criteria of the CPR and the MSDS contains all the information required by the CPR.

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# OTHER INFORMATION

NFPA, NPCA-HMIS

NPCA-HMIS Rating

Health : 1 Flammability : 0 Reactivity : 1

Personal Protection rating to be supplied by user depending on use conditions.

The data in this Material Safety Data Sheet relates only to the specific material designated herein and does not relate to use in combination with any other material or in any process.

Responsibility for MSDS : FLUOROPRODUCTS
Address : DuPont Canada Inc.

Box 2200, Streetsville,

Mississauga, Ontarion, L5M 2H3


# FREON 502

Telephone : (905)821-5935

# Indicates updated section.

End of MSDS

#### FREON

\* Canadian Centre for Occupational Health and Safety \* \* \* \* \* \* \* \* \* \* \* \* \* Issue : 2001-1 (February, 2001) \*

\*\*\* IDENTIFICATION \*\*\*

MSDS RECORD NUMBER : 1185887
PRODUCT NAME(S) : Freon 12

PRODUCT NAME(S) : Freon 12, R 12
PRODUCT IDENTIFICATION : CAS No.: 75-71-8
Form No. F-85312-4

DATE OF MSDS : 1995-05-19

\*\*\* MANUFACTURER INFORMATION \*\*\*

MANUFACTURER : ANSUL INCORPORATED

ADDRESS : One Stanton Street

Marinette Wisconsin

U.S.A. 54143-2542 Telephone: 715-735-7411 (Other

Telephone: 713-735-7411 (OC

Information Calls)
EMERGENCY TELEPHONE NO.: 800-424-9300 (CHEMTREC)

FREON 12, R 12

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QUICK IDENTIFIER (In Plant Common Name)

Prepared By: Safety and Health Department

Date Prepared: May 19, 1995

SECTION 1 - IDENTITY

Common Name: (used on label) Freon 12, R 12

(Trade Name and Synonyms)

CAS No.: 75-71-8

Chemical Dichlorodifluoromethane Chemical Halogenated Methane

Name: Family:

Formula: CCl2F

FREON SECTION 2 - INGREDIENTS PART A - HAZARDOUS INGREDIENTS ------Principal Hazardous Component(s) (chemical and common name(s)): Wt. % CAS No. \_\_\_\_\_ Dichlorodifluoromethane 75-71-8 ACGIH TLV: 1,000 ppm Acute Toxicity Data: LC50(rat) 800,000 ppm/30 min. \_\_\_\_\_\_ PART B - OTHER INGREDIENTS \_\_\_\_\_\_ Other Component(s) (chemical and common name(s)): Wt. % CAS No. N/A Acute Toxicity Data: N/A \_\_\_\_\_\_ SECTION 3 - PHYSICAL AND CHEMICAL CHARACTERISTICS (Fire and Explosion Data) \_\_\_\_\_\_ Boiling Point: -21.6 deg F Specific Gravity (H2O=1): 1.33 Vapor Pressure (mm Hg): 70.1 psi @ 70 deg F Percent Volatile by Volume (%): 100 Vapor Density (Air = 1): 4.3 Evaporation Rate ( = 1): N/A Gas at room temperature Solubility in Water: Negligible Reactivity in Water: Unreactive Appearance and Odor: Colorless gas, sweet odor. Flash Point: None Flammable Limits in Air % by Volume: N/A Extinguisher Media: N/A Auto-Ignition Temperature: N/A Use water to cool fire-exposed cylinders or other Special Fire

containers. Self-contained breathing apparatus with Fighting Procedures: full facepiece and protective clothing when re-

entering unventilated fire areas where product has

been used.

Unusual Fire and Containers are equipped with pressure and temperature

Explosion Hazards: relief devices, but rupture may occur under fire

conditions and toxic decomposition by-products may be

formed if used in fires over 900 deg F.

# FREON

SECTION 4 - PHYSICAL HAZARDS

Stability: Unstable [ ] Conditions Can be decomposed under fire Stable [X] to Avoid: conditions above 900 deg F.

Incompatibility

Active metals and fires involving metal hydrides.

(Materials to Avoid):

Hazardous

Thermal decomposition at temperatures above 900 deg F Decomposition Products: forming hydrochloric and hydrofluoric acids. These

by-products have a sharp irritating odor. They are dangerous even in low concentrations, and in sufficient concentrations can result in personal

injury or death.

Hazardous May Occur [] Conditions N/A Polymerization: Will Not Occur [X] to Avoid:

NOTE: As used in Ansul extinguishers or cylinders, Freon 12 is a gas

compressed under pressure up to 360 psi at 70 deg F.

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# SECTION 5 - HEALTH HAZARDS

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Threshold 1000 ppm is the OSHA PEL and the ACGIH TLV. NOTE: The effects of exposure to Freon 12 should disappear quickly Limit Value: upon removal from exposure. LC50 rats greater than

800,000 ppm/30 min.

Routes of Entry:

The liquid form of this material can produce chilling Eye Contact:

sensations and discomfort, also frostbite.

Skin Contact: Evaporation of liquid from the skin can produce chilling

sensations. Frostbite can occur.

Vapor is heavier than air and can cause suffocation by Inhalation:

reducing oxygen available for breathing. Breathing very high concentrations of vapor can cause lightheadedness, giddiness, shortness of breath, and may lead to narcosis, cardiac irregularities, unconsciousness or even death.

LC50 rats, 800,000 ppm/30 min.

Ingestion is not likely to occur since this material is Ingestion:

gas at room temperature.

Signs and Symptoms:

Dizziness, impaired coordination, reduced mental Acute Overexposure:

acuity, and cardiac effects can occur.

Unconsciousness or even death in high concentrations

with longer exposures.

Chronic Overexposure: None known when occupational exposures are below the

TLV.

Medical Conditions Generally: Cardiac problems.

\_\_\_\_\_\_\_\_\_\_\_\_ FREON \_\_\_\_\_\_\_\_\_\_\_\_\_\_ Aggravated by Exposure: Chemical Listed as Carcinogen or Potential: National Toxicology Yes [] I.A.R.C. Yes [] OSHA: Yes [] Program: No [X] Monographs: No [X] No [X]

# SECTION 6 - EMERGENCY AND FIRST AID PROCEDURES

Immediately flush eyes with plenty of water for at least Eye Contact: 15 minutes while holding lids open. If redness, itching or a burning sensation develops, get Medical attention.

Treat for frostbite if necessary.

Wash the material off the skin with copious amounts of Skin Contact:

soap and water for at least 15 minutes. If redness, itching, or burning occurs, get Medical attention. Treat

for frostbite if necessary.

Remove victim to fresh air. If cough or other respiratory Inhalation:

symptoms occur, consult Medical personnel. If not

breathing, give artificial respiration, preferably mouth-

to-mouth. If breathing is difficult, give oxygen.

Consult Medical personnel.

If patient is conscious, give 1 to 2 glasses of warm water Ingestion:

to drink and get Medical attention. DO NOT INDUCE

VOMITING. Have victim lie down and keep warm.

NOTE TO PHYSICIAN: Product is an asphxiant and can induce cardiac muscle

sensitization to circulating epinephrine-like compounds. Do NOT give adrenalin or similar sympathomimetic drugs. Do NOT allow victim to exercise until 24 hours following specific exposures. Freeze burns of mucosal tissue can

develop following specific exposures.

# SECTION 7 - SPECIAL PROTECTION INFORMATION

\_\_\_\_\_\_\_ Respiratory Protection Not normally necessary if controls are adequate.

(Specify Type): Self-contained breathing apparatus must be worn when using this product in testing Halon suppression

systems.

Ventilation: Local Exhaust: Mechanical (General):

Recommended in low areas or indoors where vapors may collect. Recommended to control exposures.

See mechanical.

Protective Lined butyl gloves for Eye Chemical goggles Protection: recommended. Full Gloves: handling liquid.

faceshield in addition if splashing of liquid form

# FREON

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is possible.

Other Protective Eye wash and safety showers are good safety practice Clothing or Equipment: in work areas when working with liquefied product.

\_\_\_\_\_

# SECTION 8 - SPECIAL PRECAUTIONS AND SPILL/LEAK PROCEDURES

Precautions to be Taken Store as a liquefied compressed gas in DOT approved pressure vessels away from high in Handling and Storage: temperatures. If cylinder is not attached to a system, it must be safety capped to protect against actuation of valve and

release of agent.

Other Note incompatibility information in Section

Precautions:

Steps to be Taken in Case Evacuate area; ventilate to outside

Material is Released or Spilled: atmosphere. Cool or remove hot, metal surfaces or source of non-extinguished

flames.

Waste Disposal EPA Hazardous Waste No. UO 75. Dispose of

in compliance with local, state, and Methods:

federal regulations.

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# HAZARDOUS MATERIAL IDENTIFICATION SYSTEM RATINGS

# HAZARD INDEX:

4 Severe Hazard

1 HEALTH 0 FLAMMABILITY 0 REACTIVITY 3 Serious Hazard 2 Moderate Hazard

1 Slight Hazard 0 Minimal Hazard

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N/A = Not Applicable NDA = No Data Available

ANSUL is a registered trademark

Form No. F-85312-4

# GASOLINE (GENERIC)

MSDS

\* Canadian Centre for Occupational Health and Safety \* \* \* \* \* \* \* \* \* \* \* \* \* \* Issue : 2001-1 (February, 2001) \*

\*\*\* IDENTIFICATION \*\*\*

MSDS RECORD NUMBER : 2461632

PRODUCT NAME (S) : GASOLINE (GENERIC) PRODUCT IDENTIFICATION : MSDS Number: 002914

: 2000-07-22 DATE OF MSDS

CURRENCY NOTE : This MSDS was provided to CCOHS in

electronic form on 2000-10-03

\*\*\* MANUFACTURER INFORMATION \*\*\*

MANUFACTURER : Chevron Products Company

ADDRESS : 6001 Bollinger Canyon Road

San Ramon California

U.S.A. 94583

Telephone: 800-689-3998 (Product

Information, MSDS Requests) 510-242-5357 (Product Information, Technical Information)

EMERGENCY TELEPHONE NO.: 800-231-0623 (Health, 24 hr)

510-231-0623 (International, Health, 24

hr)

800-424-9300 (CHEMTREC, Transportation,

24 hr)

703-527-3887 (Transportation 24hr,

Emergency Info Centers are in USA, Int'l

collect calls accepted)

# 1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

GASOLINE (GENERIC)

EMERGENCY TELEPHONE NUMBERS COMPANY IDENTIFICATION

Chevron Products Company Marketing, MSDS Coordinator 6001 Bollinger Canyon Road

San Ramon, CA 94583

(800)424-9300 or (703)527-3887 Emergency Information Centers are located in U.S.A. Int'l collect calls accepted

HEALTH (24 hr): (800)231-0623 or

(510)231-0623 (International) TRANSPORTATION (24 hr): CHEMTREC

PRODUCT INFORMATION: (800)689-3998 MSDS Requests and Product Information

# GASOLINE (GENERIC)

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# 2. COMPOSITION/INFORMATION ON INGREDIENTS

100	0	0.	GASOLINE	/ CIDNIDD TO

an	BTITT	T	STT	BTC
CO	IN.T.	AL	IN T	NG

COMPONENTS	AMOUNT	LIMIT/QTY	AGENCY/TYPE
GASOLINE (GENERIC)			
	100.00%	890 mg/m3 1480 mg/m3	ACGIH TWA ACGIH STEL
		2000 mg/m3	OSHA PEL

# POTENTIALLY

INCLUDING

BENZENE

Chemical Name: BENZENE

CAS71432 < 5.00% 0.5 ppm ACGIH TWA

Revision Number: 14 Revision Date: 07/22/00 MSDS Number: 002914

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Page 2 of 15 GASOLINE (GENERIC)

> ACGIH STEL 2.5 ppm 1 ppm OSHA PEL OSHA CEILING 5 ppm 10 LBS CERCLA 302.4 RQ

ETHYL BENZENE

Chemical Name: BENZENE, ETHYL-

ACGIH TWA 100 ppm CAS100414 ACGIH STEL 125 ppm OSHA PEL 100 ppm

CERCLA 302.4 RQ 1,000 LBS

XYLENE

Chemical Name: BENZENE, DIMETHYL-

ACGIH TWA ACGIH STEL 100 ppm CAS1330207 150 ppm OSHA PEL 100 ppm

CERCLA 302.4 RQ 100 LBS

TOLUENE

Chemical Name: TOLUENE

CAS108883 50 ppm ACGIH TWA 200 ppm OSHA PEL

	(GENERIC)	
	300 ppm 1,000 LBS	OSHA CEILING CERCLA 302.4 RQ
N-BUTANE Chemical Name: N-BUTANE CAS106978	800 ppm	ACGIH TWA
N-HEPTANE Chemical Name: N-HEPTANE CAS142825	400 ppm 500 ppm	ACGIH TWA ACGIH STEL
	500 ppm	OSHA PEL
N-HEXANE Chemical Name: N-HEXANE CAS110543	50 ppm 500 ppm	ACGIH TWA
	5,000 LBS	CERCLA 302.4 RQ
HEXANE ISOMERS (OTHER THAN N) HEXANES		
	500 ppm 1000 ppm	ACGIH TWA ACGIH STEL
PENTANE (ALL ISOMERS) PENTANES		
	600 ppm 750 ppm 1000 ppm	ACGIH TWA ACGIH STEL OSHA PEL
CYCLOHEXANE Chemical Name: CYCLOHEXANE		
CAS110827	300 ppm 300 ppm 1,000 LBS	ACGIH TWA OSHA PEL CERCLA 302.4 RQ
METHYLCYCLOHEXANE Chemical Name: CYCLOHEXANE, METHYL	274	
CAS108872	400 ppm 500 ppm	ACGIH TWA OSHA PEL
TRIMETHYLBENZENE		
Chemical Name: BENZENE, TRIMETHYL- CAS25551137	25 ppm	ACGIH TWA
2,2,4-TRIMETHYLPENTANE Chemical Name: 2,2,4-TRIMETHYLPENTANE CAS540841	1,000 LBS	CERCLA 302.4 RQ
CAN CONTAIN		
METHYL TERT BUTYL ETHER (MTBE) Chemical Name: 2-METHOXY-2-METHYL PROPA	NE	
Revision Number: 14 Revision Date	: 07/22/00 MS	DS Number: 002914
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# GASOLINE (GENERIC)

CAS1634044 < 15.00% 40 ppm ACGIH TWA 50 ppm Chevron STEL

1,000 LBS CERCLA 302.4 RO

ETHYL TERT BUTYL ETHER (ETBE)

Chemical Name: 2-ETHOXY-2-METHYL PROPANE

CAS637923 < 18.00% NONE NA

TERT-AMYL METHYL ETHER (TAME)

Chemical Name: 2-METHOXY-2-METHYL-BUTANE

CAS994058 < 17.00% 50 ppm Chevron STEL

OR

ETHANOL

Chemical Name: ETHYL ALCOHOL

CAS64175 < 10.00% 1000 ppm ACGIH TWA 1000 ppm OSHA PEL

# COMPOSITION COMMENT:

Refer to the OSHA Benzene Standard (29 CFR 1910.1028) and Table Z-2 for detailed training, exposure monitoring, respiratory protection and medical surveillance requirements before using this product.

Motor gasoline is considered a mixture by EPA under the Toxic Substances Control Act (TSCA). The refinery streams used to blend motor gasoline are all on the TSCA Chemical Substances Inventory. The appropriate CAS number for refinery blended motor gasoline is 86290-81-5. The product specifications of motor gasoline sold in your area will depend on applicable Federal and State regulations. Ethyl Alcohol is only added in limited specific distribution areas.

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# 3. HAZARDS IDENTIFICATION

Variable colored liquid with a petroleum hydrocarbon odor.

- EXTREMELY FLAMMABLE
- HARMFUL OR FATAL IF SWALLOWED CAN ENTER LUNGS AND CAUSE DAMAGE
- VAPOR HARMFUL

# GASOLINE (GENERIC)

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- MAY CAUSE EYE AND SKIN IRRITATION
- LONG-TERM EXPOSURE TO VAPOR HAS CAUSED CANCER IN LABORATORY ANIMALS
- KEEP OUT OF REACH OF CHILDREN

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# IMMEDIATE HEALTH EFFECTS

#### EYE

Contact with the eyes causes irritation. Eye contact with the vapors, fumes, or spray mist from this substance could also cause similar signs and symptoms.

# SKIN:

Contact with the skin causes irritation. Not expected to be harmful to internal organs if absorbed through the skin. Prolonged or frequently repeated contact may cause the skin to become cracked or dry from the defatting action of this material.

#### INGESTION:

Because of the low viscosity of this substance, it can directly enter the lungs if it is swallowed (this is called aspiration). This can occur during the act of swallowing or when vomiting the substance. Once in the lungs, the substance is very difficult to remove and can cause severe injury to the lungs and death.

# INHALATION:

May be harmful if inhaled. Breathing the vapors at concentrations above the recommended exposure standard can cause central nervous system effects. The vapor or fumes from this material may cause respiratory irritation.

# SIGNS AND SYMPTOMS OF EXPOSURE:

Eye damage or irritation: may include pain, tearing, reddening, swelling, and impaired vision. Skin injury: may include pain, discoloration, swelling, and blistering. Respiratory irritation: may include coughing and difficulty breathing. Central nervous system effects may include headache, dizziness, nausea, vomiting, weakness, loss of coordination, blurred vision, drowsiness, confusion, or disorientation. At extreme exposures, central nervous system effects may include respiratory depression, tremors or convulsions, loss of consciousness, coma or death.

# CARCINOGENICITY:

Risk depends on duration and level of exposure. See Section 11 for additional information. Gasoline has been classified as a Group 2B carcinogen (possibly carcinogenic to humans) by the International Agency for Research on Cancer (IARC). Contains chemical(s) known to the State of California to cause cancer. Contains benzene, which has been classified as a carcinogen by the National Toxicology Program (NTP), and a Group 1 carcinogen (carcinogenic to humans) by the International Agency for Research on Cancer (IARC). Contains ethylbenzene which has been classified as a Group 2B carcinogen (possibly carcinogenic to humans) by the International Agency for Research on

# GASOLINE (GENERIC)

# Cancer (IARC).

Whole gasoline exhaust was reviewed by the International Agency for Research on Cancer (IARC) in their Monograph Volume 46 (1989). Evidence for causing cancer was considered inadequate in animals and inadequate in humans. IARC placed whole gasoline exhaust in Category 2B, considering it possibly carcinogenic to humans.

# 4. FIRST AID MEASURES

#### EVE.

Flush eyes with water immediately while holding the eyelids open. Remove contact lenses, if worn, after initial flushing, and continue flushing for at least 15 minutes. Get medical attention if irritation persists.

# SKIN:

Wash skin immediately with soap and water and remove contaminated clothing and shoes. Get medical attention if irritation persists. Discard contaminated clothing and shoes or thoroughly clean before reuse.

# INGESTION:

If swallowed, give water or milk to drink and telephone for medical advice. DO NOT make person vomit unless directed to do so by medical personnel. If medical advice cannot be obtained, then take the person and product container to the nearest medical emergency treatment center or hospital.

# INHALATION:

Move the exposed person to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention if breathing difficulties continue.

NOTE TO PHYSICIANS:

Ingestion of this product or subsequent vomiting can result in aspiration of light hydrocarbon liquid which can cause pneumonitis.

# FIRE FIGHTING MEASURES

# FIRE CLASSIFICATION:

OSHA Classification (29 CFR 1910.1200): Flammable liquid. See section 7 for appropriate handling and storage conditions. FLAMMABLE PROPERTIES:

FLASH POINT: (TCC) < -49F (<-45C)

AUTOIGNITION: 536F (280C)

FLAMMABILITY LIMITS (% by volume in air): Lower: 1.4 Upper: 7.6

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EXTINGUISHING MEDIA:

CO2, Dry Chemical, Fire Fighting Foam, AFFF.
NFPA RATINGS: Health 1; Flammability 3; Reactivity 0.

# FIRE FIGHTING INSTRUCTIONS:

Use water spray to cool fire-exposed containers and to protect personnel. For fires involving this material, do not enter any enclosed or confined fire space without proper protective equipment, including self-contained breathing apparatus.

# COMBUSTION PRODUCTS:

Normal combustion forms carbon dioxide and water vapor; incomplete combustion can produce carbon monoxide.

# 6. ACCIDENTAL RELEASE MEASURES

CHEMTREC EMERGENCY NUMBER (24 hr): (800)424-9300 or (703)527-3887 International Collect Calls Accepted

# ACCIDENTAL RELEASE MEASURES:

Eliminate all sources of ignition in the vicinity of the spill or released vapor. Stop the source of the leak or release. Clean up releases as soon as possible, observing precautions in Exposure Controls/Personal Protection. Contain liquid to prevent further contamination of soil, surface water or groundwater. Clean up small spills using appropriate techniques such as sorbent materials or pumping. Where feasible and appropriate, remove contaminated soil. Follow prescribed procedures for reporting and responding to larger releases. Place contaminated materials in disposable containers and dispose of in a manner consistent with applicable regulations. Contact local environmental or health authorities for approved disposal of this material.

Release of this product should be prevented from contaminating soil and water and from entering drainage and sewer systems. U.S.A. regulations require reporting spills of this material that could reach any surface waters. The toll free number for the U.S. Coast Guard National Response Center is (800) 424-8802.

# 7. HANDLING AND STORAGE

This product presents an extreme fire hazard. Liquid very quickly evaporates, even at low temperatures, and forms vapor (fumes) which can catch fire and burn with explosive violence. Invisible vapor spreads easily and can be set on fire by many sources such as pilot lights,

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welding equipment, and electrical motors and switches.

Electrostatic charge may accumulate and create a hazardous condition when handling this material. To minimize this hazard, bonding and grounding may be necessary but may not, by themselves, be sufficient. Review all operations which have the potential of generating an accumulation of electrostatic charge and/or a flammable atmosphere (including tank and container filling, splash filling, tank cleaning, sampling, gauging, switch loading, filtering, mixing, agitation, and vacuum truck operations) and use appropriate mitigating procedures. For more information, refer to OSHA Standard 29 CFR 1910.106, "Flammable and Combustible Liquids", National Fire Protection Association (NFPA) 77, "Recommended Practice on Static Electricity", and/or the American Petroleum Institute (API) Recommended Practice 2003, "Protection Against Ignitions Arising Out of Static, Lightning, and Stray Currents".

Improper filling of portable gasoline containers creates danger of fire. Only dispense gasoline into approved and properly labeled gasoline containers. Always place portable containers on the ground. Be sure pump nozzle is in contact with the container while filling. Do not use a nozzle's lock-open device. Do not fill portable containers that are inside a vehicle or truck/trailer bed.

Never siphon gasoline by mouth. Use only as a motor fuel. Do not use for cleaning, pressure appliance fuel, or any other such use. DO NOT USE OR STORE near heat, sparks or open flames. USE AND STORE ONLY IN WELL VENTILATED AREA. Keep container closed when not in use. READ AND OBSERVE ALL PRECAUTIONS ON PRODUCT LABEL.

Container is not designed to contain pressure. Do not use pressure to empty container or it may rupture with explosive force. Empty containers retain product residue (solid, liquid, and/or vapor) and can be dangerous. Do not pressurize, cut, weld, braze, solder, drill, grind, or expose such containers to heat, flame, sparks, static electricity, or other sources of ignition. They may explode and cause injury or death. Empty containers should be completely drained, properly closed, and promptly returned to a drum reconditioner, or properly disposed of.

Do not get in eyes, on skin, or on clothing. Do not taste or swallow. Do not breathe vapor or fumes. Do not breathe mist. Wash thoroughly after handling.

# 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

# GENERAL CONSIDERATIONS:

Consider the potential hazards of this material (see Section 3), applicable exposure limits, job activities, and other substances in the work place when designing engineering controls and selecting personal protective equipment. If engineering controls or work practices are not adequate to prevent exposure to harmful levels of this material, the personal protective equipment listed below is recommended. The user should

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read and understand all instructions and limitations supplied with the equipment since protection is usually provided for a limited time or under certain circumstances.

# ENGINEERING CONTROLS

Use process enclosures, local exhaust ventilation, or other engineering controls to control airborne levels below the recommended exposure limits.

# PERSONAL PROTECTIVE EQUIPMENT

EYE/FACE PROTECTION:

No special eye protection is normally required. Where splashing is possible, wear safety glasses with side shields as a good safety practice.

# SKIN PROTECTION:

No special protective clothing is normally required. Where splashing is possible, select protective clothing depending on operations conducted,

physical requirements and other substances. Suggested materials for protective gloves include: <Nitrile> <Polyurethane> <Viton> <Chlorinated Polyethylene (or Chlorosulfonated Polyethylene or CPE)>

# RESPIRATORY PROTECTION:

Determine if airborne concentrations are below the recommended exposure limits. If not, wear a NIOSH approved respirator that provides adequate protection from measured concentrations of this material. Use the following respirators: Organic Vapor. Use a positive pressure, air-supplying respirator if there is potential for uncontrolled release, exposure levels are not known, or other circumstances where air-purifying respirators may not provide adequate protection.

# 9. PHYSICAL AND CHEMICAL PROPERTIES

# PHYSICAL DESCRIPTION:

Variable colored liquid with a petroleum hydrocarbon odor.

pH: N.

VAPOR PRESSURE: 5 - 15 PSI @ 100F (REID)

VAPOR DENSITY

(AIR=1): 3-4

BOILING POINT: 25 - 225C (range)

FREEZING POINT: NA
MELTING POINT: NA

SOLUBILITY: Soluble in hydrocarbons; insoluble in water.

SPECIFIC GRAVITY: 0.7 - 0.8 @ 15.6/15.6C PERCENT VOLATILE (VOL): 99+%

# GASOLINE (GENERIC)

# 10. STABILITY AND REACTIVITY

HAZARDOUS DECOMPOSITION PRODUCTS:

None known

CHEMICAL STABILITY:

Stable.

CONDITIONS TO AVOID:

See section 7.

INCOMPATIBILITY WITH OTHER MATERIALS:

May react with strong oxidizing agents, such as chlorates, nitrates, peroxides, etc.

HAZARDOUS POLYMERIZATION:

Polymerization will not occur.

# 11. TOXICOLOGICAL INFORMATION

# EYE EFFECTS:

The mean 24-hour Draize eye irritation score in rabbits is 0.

# SKIN EFFECTS:

This material was not a skin sensitizer in the modified Buehler Guinea Pig Sensitization Test. For a 4-hour exposure, the Primary Irritation Index (PII) in rabbits is: 4.8.

# ACUTE ORAL EFFECTS:

The oral LD50 in rats is > 5 ml/kg.

# ACUTE INHALATION EFFECTS:

No product toxicology data available.

# ADDITIONAL TOXICOLOGY INFORMATION:

When vapor exposures are low, or short duration and infrequent, such as during refuelling and tanker loading/unloading, neither total hydrocarbon nor components such as benzene are likely to result in any adverse health effects. In situations such as accidents or spills where exposure to gasoline vapor and liquid is potentially high, attention should be paid to potential toxic effects of specific components in addition to those of total hydrocarbons. Information about specific components in gasoline are found in Section 1 and Section 15 of this MSDS. More detailed information on the health hazard of specific gasoline components can be obtain from the Chevron Emergency Information Center (see Section 1 for telephone numbers).

A study was done in which ten volunteers were exposed for 30 minutes to about 200, 500 or 1000 ppm concentrations of the vapor of three different Revision Number: 14 Revision Date: 07/22/00 MSDS Number: 002914