Appendix "G"

♦ Fuel MSDS Sheets ♦

322-110

Revision Number: 1



Shell Canada Limited Material Safety Data Sheet

Effective Date: 19980901





Class B3 Combustible Class D2B Other Toxic Liquid Effects - Skin Irritant

1. PRODUCT AND COMPANY IDENTIFICATION

PRODUCT:

REGULAR SULPHUR DIESEL FUEL

SYNONYMS:

Diesel

Automotive Gas Oil

PRODUCT USE:

Fuel Solvent

MSDS Number:

322-110

MANUFACTURER

TELEPHONE NUMBERS

Shell Canada Limited

1-800-661-7378 Shell Emergency Number

P.O. Box 100, Station M 400-4th Ave. S.W.

CANUTEC 24 HOUR EMERGENCY NUMBER 613-996-6666

Calgary, AB Canada

For general information: 1-800-661-1600 403-691-3982

T2P 2H5

For MSDS information:

(From 7:30 to 4:30 Mountain Time)

403-691-2220

This MSDS was prepared by the Toxicology and Material Safety Section of Shell Canada Limited.

2. COMPOSITION/INFORMATION ON INGREDIENTS

Component Name	CAS	%	WHMIS	CBI Claim No.
	Number	Range	Controlled	CBI Date
REGULAR SULPHUR DIESEL FUEL	68476-34-6	100	Yes	

See Section 8 for Occupational Exposure Guidelines.

3. HAZARDS IDENTIFICATION

Physical Description: Liquid Lightly Coloured Hydrocarbon Odour

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REGULAR SULPHUR DIESEL FUEL

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Routes of Exposure: Exposure may occur via inhalation, ingestion, skin absorption and skin or eye

contact.

Hazards:

Combustible Liquid. Irritating to skin.

Vapours are moderately irritating to the eyes.

Vapours are moderately irritating to the respiratory passages. The liquid when accidently aspirated into the lungs can cause a severe inflammation of the

lung.

Handling: Eliminate all ignilion sources.

Avoid prolonged exposure to vapours. Wear suitable gloves and eye protection.

Bond and ground transfer containers and equipment to avoid static

accumulation.

Empty containers are hazardous, may contain flammable / explosive dusts.

liquid residue or vapours. Keep away from sparks and open flames.

For further information on health effects, see Section 11.

4. FIRST AID

Eyes Flush eyes with water for at least 15 minutes while holding eyelids open. If

irritation occurs and persists, obtain medical attention.

Skin Flush affected skin with gently flowing lukewarm water for at least 20 minutes

and remove contaminated clothing while rinsing. Wash contaminated skin with mild soap and water for 15 minutes. If irritation occurs and persists, obtain

medical attention.

Ingestion DO NOT INDUCE VOMITING! OBTAIN MEDICAL ATTENTION IMMEDIATELY.

Guard against aspiration into lungs by having the individual turn on to their left side. If vomiting occurs spontaneously keep head below hips to prevent aspiration of liquid into the lungs. Do not give anything by mouth to an

unconscious person.

Inhalation Remove victim from further exposure and restore breathing, if required. Obtain

medical attention.

Notes to Physician The main hazard following accidental ingestion is aspiration of the liquid into the

lungs producing chemical pneumonitis. If more than 2.0 mL/kg has been

ingested, vomiting should be induced with supervision. If symptoms such as loss of gag reflex, convulsions or unconsciousness occur before vomiting, gastric

lavage with a cuffed endotracheal tube should be considered.

5. FIRE FIGHTING MEASURES

Extinguishing Media

Dry Chemical Carbon Dioxide

Foam Water Fog

REGULAR SULPHUR DIESEL FUEL

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Revision Number: 1

Firefighting Instructions

Vapour forms a flammable/explosive mixture with air between upper and lower flammable limits. Vapours may travel along ground and flashback along vapour trail may occur. Do not use water except as a fog. Product will float and can be reignited on surface of water. Containers exposed to intense heat from fires should be cooled with water to prevent vapour pressure buildup which could result in container rupture. Container areas exposed to direct flame contact should be cooled with large quantities of water as needed to prevent weakening of container structure. Caution - Combustible. Do not enter confined fire space without adequate protective clothing and an approved positive pressure self-contained breathing apparatus. A complex mixture of airborne solid, liquid, particulates and gases will evolve when this material undergoes pyrolysis or combustion. Carbon dioxide, carbon monoxide and unidentified organic

Hazardous Combustion **Products**

6. ACCIDENTAL RELEASE MEASURES

Issue warning "Combustible". Eliminate all ignition sources. Isolate hazard area and restrict access. Handling equipment must be grounded. Try to work upwind of spill. Avoid direct contact with material. Wear appropriate breathing apparatus (if applicable) and protective clothing. Stop leak only if safe to do so. Dike and contain land spills; contain water spills by booming. Use water fog to knock down vapours; contain runoff. Absorb residue or small spills with absorbent material and remove to non-leaking containers for disposal. Recommended materials: Clay or Sand Flush area with water to remove trace residue. Dispose of recovered material as noted under Disposal Considerations. Notify appropriate environmental agency(ies).

compounds may be formed upon combustion.

7. HANDLING AND STORAGE

Handling:

Combustible. Avoid excessive heat, sparks, open flames and all other sources of ignition. Fixed equipment as well as transfer containers and equipment should be grounded to prevent accumulation of static charge. Vapours are heavier than air and will settle and collect in low areas and pits, displacing breathing air. Extinguish pilot lights, cigarettes and turn off other sources of ignition prior to use and until all vapours are gone. Vapours may accumulate and travel to distant ignition sources and flashback. Do not cut, drill, grind, weld or perform similar operations on or near containers. Empty containers are hazardous, may contain flammable/explosive dusts, residues or vapours. Do not pressurize drum containers to empty them. Never siphon by mouth. Wash with soap and water prior to eating, drinking, smoking, applying cosmetics or using toilet facilities. Launder contaminated clothing prior to reuse. Use good personal hygiene.

Storage:

Use explosion-proof ventilation to prevent vapour accumulation. Keep container tightly closed.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

THE FOLLOWING INFORMATION, WHILE APPROPRIATE FOR THIS PRODUCT, IS GENERAL IN NATURE. THE SELECTION OF PERSONAL PROTECTIVE EQUIPMENT WILL VARY DEPENDING ON THE CONDITIONS OF USE.

REGULAR SULPHUR DIESEL FUEL

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Occupational Exposure Limits (1998): North American exposure limits have not been established

for the product. Consult local authorities for acceptable

provincial values.

Oil mist (mineral): 5 mg/m3 (TLV/TWA) ACGIH

10 mg/m3 (TLV/STEL) ACGIH

Recommend SHELL guideline of 125 mg/m3 for vapours (8

hour shift).

Mechanical Ventilation: Use explosion-proof ventilation as required to control vapour

concentrations. Concentrations in air should be maintained below lower explosive limit at all times or below the recommended threshold limit value if unprotected personnel are involved. Make up air should always be supplied to balance air exhausted (either generally or locally). For personnel entry into confined spaces (i.e. bulk storage tanks) a proper confined space entry procedure must be followed including ventilation

and testing of tank atmosphere.

PERSONAL PROTECTIVE EQUIPMENT:

Eye Protection: Chemical safety goggles and/or full face shield to protect eyes and

face, if product is handled such that it could be splashed into eyes.

Provide an eyewash station in the area.

Skin Protection: Impervious gloves (viton, nitrile) should be worn at all times when

handling this material. In confined spaces or where the risk of skin exposure is much higher, impervious clothing should be worn. Safety

showers should be available for emergency use.

Respiratory Protection: If exposure exceeds occupational exposure limits, use an appropriate

NIOSH-approved respirator. Use a NIOSH-approved chemical cartridge respirator with organic vapour cartridges or use a NIOSH-approved supplied-air respirator. For high airborne concentrations, use a NIOSH-

approved supplied air respirator, eitner self-contained or airline breathing apparatus, operated in positive pressure mode.

9. PHYSICAL DATA

Physical State: Liquid

Appearance: Lightly Coloured
Odour: Hydrocarbon Odour

Odour Threshold: Not available Freezing/Pour Point: Not available

Boiling Point: 246 - 388 degrees C

Density: <876 kg/m3 @ 15 degrees C

Vapour Density (Air = 1): Not available
Vapour Pressure: Not available
pH: Not applicable

Flash Point: Method Pensky-Martens CC >40 degrees C

Lower Explosion Limit: 1 % (vol.)
Upper Explosion Limit: 6 % (vol.)
Autoignition Temperature: 250 degrees C

Viscosity: 1.3 - 4.1 cSt @ 40 degrees C

Evaporation Rate (n-BuAc = 1): Not available Partition Coefficient (Kow): Not available Water Solubility: Insoluble

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Other Solvents:

Formula:

Hydrocarbon Solvents C10 to C22 Hydrocarbons

10, STABILITY AND REACTIVITY

Chemically Stable: Yes
Hazardous Polymerization: No
Sensitive to Mechanical Impact: No

Sensitive to Static Discharge: Yes

Hazardous Decomposition Products: Thermal decomposition products are highly dependent on

combustion conditions.

Incompatible Materials:

Avoid strong oxidizing agents.

Conditions of Reactivity:

Avoid excessive heat, open flames and all ignition

sources.

11. TOXICOLOGICAL INFORMATION

Ingredient (or Product if not specified)

Toxicological Data

REGULAR SULPHUR DIESEL FUEL

LD50 Oral Rat >5000 mg/kg LD50 Dermal Rabbit >2000 mg/kg

Routes of Exposure:

Exposure may occur via inhalation, ingestion, skin absorption and skin or

eye contact.

Irritancy:

This product is expected to be irritating to skin but is not predicted to be a

skin sensitizer.

Chronic Effects:

Prolonged and repeated contact with skin can cause defatting and drying of the skin resulting in skin irritation and dermatitis. Prolonged exposure to high vapour concentration can cause headache, dizziness, nausea, blurred vision and central nervous system depression. Prolonged or repeated contact may cause various forms of dermatitis including folliculitis and oil acne.

Pre-existing Conditions:

Pre-existing eye, skin and respiratory disorders may be aggravated by

exposure to this product,

Carcinogenicity and

Mutagenicity:

The International Agency for Research on Cancer (IARC) considers that this product is not classifiable as to its carcinogenicity to humans. Midgle distillates have caused skin cancers in laboratory animals when applied repeatedly and left in place between applications. This effect is believed

to be caused by the continuous irritation of the skin. Good personal

hygiene should be maintained to avoid this risk.

12. ECOLOGICAL INFORMATION

Environmental Effects Do not allow product or runoff from fire control to enter storm or sanitary

sewers, lakes, rivers, streams, or public waterways. Block off drains and ditches. Provincial regulations require and federal regulations may require that environmental and/or other agencies be notified of a spill incident. Spill area must be cleaned and restored to original condition or to the satisfaction of authorities. May cause physical fouling of aquatic

organisms.

Biodegradability Not read

Not readily biodegradable. Potential for bioaccumulation.

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13. DISPOSAL CONSIDERATIONS

Waste management priorities (depending on volumes and concentration of waste) are: 1. recycle (reprocess), 2. energy recovery (cement kilns, thermal power generation), 3. incineration, 4. disposal at a licenced waste disposal facility. Do not attempt to combust waste on-site. Incinerate at a licenced waste disposal site with approval of environmental authority.

14. TRANSPORTATION INFORMATION

Canadian Road and Rail Shipping Classification:

UN/NA Number

UN1202

Proper Shipping Name

FUEL OIL

Hazard Class

Class 3 Flammable Liquid

Packing Group

PG III

Shipping Description

FUEL OIL Class 3 UN1202 PG III

15. REGULATORY INFORMATION

This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all the Information required by the Controlled Products Regulations.

WHMIS Class:

Class B3 Combustible Liquid

Class D2B Other Toxic Effects - Skin Irritant

DSL/NDSL Status:

This product, or all components, are listed on the Domestic Substances List, as required under the Canadian Environmental Protection Act.

Other Regulatory Status:

No Canadian federal standards.

16. ADDITIONAL INFORMATION

LABEL STATEMENTS

Hazard Statement:

Combustible Liquid.

Irritating to skin.

Handling Statement:

Eliminate all ignition sources.

Avoid prolonged exposure to vapours. Wear suitable gloves and eye protection.

Bond and ground transfer containers and equipment to avoid static

accumulation.

Empty containers are hazardous, may contain flammable / explosive dusts,

liquid residue or vapours. Keep away from sparks and open flames.

First Aid Statement:

Wash contaminated skin with soap and water.

Flush eyes with water.

If overcome by vapours remove to fresh air.

Do not induce vomiting. Obtain medical attention. REGULAR SULPHUR DIE L FUEL

322-110

Revision Number: 1

Revisions:

This MSDS has been reissued in the ANSI Z400.1 standard format.

SHELL JET B

141-012

Revision Number: 5



Shell Canada Limited Material Safety Data Sheet

Effective Date: 19971203







Liquid

Effects - Skin Irritant

Class B2 Flammable Class D2B Other Toxic Class D2A Other Toxic Effects - Carcinogen

1. PRODUCT AND COMPANY IDENTIFICATION

PRODUCT:

SHELL JET B

SYNONYMS:

WIDE BOILING RANGE AVIATION TURBINE FUEL

PRODUCT USE:

Fuel

MSDS Number:

141-012

MANUFACTURER Shell Canada Limited **TELEPHONE NUMBERS** Shell Emergency Number

1-800-661-7378

P.O. Box 100, Station M 400-4th Ave. S.W.

CANUTEC 24 HOUR EMERGENCY NUMBER

613-996-6666

Calgary, AB Canada

For general information:

1-800-661-1600

T2P 2H5

For MSDS Information: (From 7:30 to 4:30 Mountain Time)

403-691-3982 403-691-2220

This MSDS was prepared by the Toxicology and Material Safety Section of Shell Canada Limited.

2. COMPOSITION/INFORMATION ON INGREDIENTS

Component Name	CAS Number	% Range	WHMIS Controlled	CBI Claim No. CBI Date
Naphtha (Petroleum), Full-range Reformed	68919-37-9	>95	Yes	
Benzene	71-43-2	0.5 - 1.5	Yes	

See Section 8 for Occupational Exposure Guidelines.

3. HAZARDS IDENTIFICATION

^{*}A star in the product name designates a trade-mark(s) of Shell Canada Limited. Used under license by Shell Canada Products Limited.

JUN 09 2000 10:43 FR SHELL CANADA - HSE 403 691 3321 TO 816049877107

SHELL JET B 141-012

Revision Number: 5

P.10/15

Physical Description: Liquid Bright Clear Typical Gasoline Odour

Routes of Exposure: Exposure may occur via inhalation, ingestion, skin absorption and skin or eye

contact.

Hazards:

Flammable Liquid. Irritating to skin. Contains Benzene. May cause cancer.

Vapours are moderately irritating to the eyes.

Vapours are moderately irritating to the respiratory passages. The liquid when accidently aspirated into the lungs can cause a severe inflammation of the

lung. Excessive exposure to benzene may cause leukemia in man.

Handling:

Eliminate all ignition sources.

Wear suitable gloves and eye protection.

Bond and ground transfer containers and equipment to avoid static

accumulation.

Avoid prolonged exposure to vapours.

Empty containers are hazardous, may contain flammable / explosive dusts,

liquid residue or vapours. Keep away from sparks and open flames.

For further information on health effects, see Section 11.

4. FIRST AID

Eyes Flush eyes with water for at least 15 minutes while holding eyelids open. If

irritation occurs and persists, obtain medical attention.

Skin Wash contaminated skin with mild soap and water for 15 minutes. If irritation

occurs and persists, obtain medical attention.

Ingestion DO NOT INDUCE VOMITING! OBTAIN MEDICAL ATTENTION IMMEDIATELY.

Guard against aspiration into lungs by having the individual turn on to their left side. If vomiting occurs spontaneously keep head below hips to prevent

aspiration of liquid into the lungs.

Inhalation Remove victim from further exposure and restore breathing, if required. Obtain

medical attention.

Notes to Physician The main hazard following accidental ingestion is aspiration of the liquid into the

lungs producing chemical pneumonitis. If more than 2.0 mL/kg has been

ingested, vomiting should be induced with supervision. If symptoms such as loss of gag reflex, convulsions or unconsciousness occur before vomiting, gastric

lavage with a cuffed endotracheal tube should be considered.

5. FIRE FIGHTING MEASURES

Extinguishing Media

Dry Chemical
Carbon Dioxide

Foam Water Fog JUN 09 2000 10:43 FR SHELL CANADA - HSE 403 691 3321 TO R16049877107 P.11/15

SHELL JET B

141-012

Revision Number: 5

Firefighting Instructions Extremely flammable. Vapour forms a flammable/explosive mixture

with air between upper and lower flammable limits. Vapours may travel along ground and flashback along vapour trail may occur. Do not use water except as a fog. Use water to cool fire exposed containers. Product will float and can be reignited on surface of water. Containers exposed to intense heat from fires should be cooled with water to prevent vapour pressure buildup which could result in container rupture. Container areas exposed to direct flame contact should be cooled with large quantities of water as needed to prevent weakening of container structure. Always stay away from ends of containers due to explosive potential. Fight fire from maximum distance. Do not enter confined fire space without adequate protective clothing and an approved positive pressure self-

contained breathing apparatus.

Hazardous Combustion Products

A complex mixture of airborne solid, liquid, particulates and gases will evolve when this material undergoes pyrolysis or combustion. Carbon dioxide, carbon monoxide and unidentified organic compounds may be formed upon combustion.

6. ACCIDENTAL RELEASE MEASURES

Issue warning "Flammable". Eliminate all ignition sources. Handling equipment must be grounded. Isolate hazard area and restrict access. Try to work upwind of spill. Avoid direct contact with material. Saturated clothing should be immediately removed to avoid flammability hazard. Wear appropriate breathing apparatus (if applicable) and protective clothing. Stop leak only if safe to do so. Dike and contain land spills; contain water spills by booming. Use water fog to knock down vapours; contain runoff. For large spills remove by mechanical means and place in containers. Absorb residue or small spills with absorbent material and remove to non-leaking containers for disposal. Recommended materials: Clay or Sand Flush area with water to remove trace residue. Dispose of recovered material as noted under Disposal Considerations. Notify appropriate environmental agency(ies).

7. HANDLING AND STORAGE

Handling: Extremely flammable. Avoid excessive heat, sparks, open flames and all other

sources of ignition. Fixed equipment as well as transfer containers and equipment should be grounded to prevent accumulation of static charge. Vapours are heavier than air and will settle and collect in low areas and pits, displacing breathing air. Extinguish pilot lights, cigarettes and turn off other sources of ignition prior to use and until all vapours are gone. Vapours may accumulate and travel to distant ignition sources and flashback. Do not cut, drill, grind, weld or perform similar operations on

or near containers. Empty containers are hazardous, may contain

flammable/explosive dusts, residues or vapours. Do not pressurize drum containers to empty them. Never siphon by mouth. Wash with soap and water prior to eating, drinking, smoking, applying cosmetics or using toilet facilities. Launder contaminated

clothing prior to reuse. Use good personal hygiene.

Storage: Use explosion-proof ventilation to prevent vapour accumulation. Keep container tightly

closed.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

403 691 3321 TO P16049877107 P.12/15 JUN 09 2000 10:44 FR SHELL CANADA - HSE

SHELL JET B 141-012 Revision Number: 5

THE FOLLOWING INFORMATION, WHILE APPROPRIATE FOR THIS PRODUCT, IS GENERAL IN NATURE. THE SELECTION OF PERSONAL PROTECTIVE EQUIPMENT WILL VARY DEPENDING ON THE CONDITIONS OF USE.

Occupational Exposure Limits (1998): North American exposure limits have not been established

for the product. Consult local authorities for acceptable

provincial values.

Gasoline: 300 ppm, 890 mg/m3 (TLV/TWA) ACGIH

500 ppm,1480 mg/m3 (TLV/STEL) ACGIH

Benzene (skin): 0.5 ppm, 1.6 mg/m3 (TLV/TWA)

2.5 ppm (STEL) ACGIH

Mechanical Ventilation: Make up air should always be supplied to balance air exhausted (either

> generally or locally). Concentrations in air should be maintained below lower explosive limit at all times or below the recommended threshold limit value if unprotected personnel are involved. For personnel entry into confined spaces (i.e. bulk storage tanks) a proper confined space entry procedure must be followed including ventilation and testing of tank atmosphere. Use explosion-proof ventilation as required to control

vapour concentrations.

PERSONAL PROTECTIVE EQUIPMENT:

Chemical safety goggles and/or full face shield to protect eyes and Eye Protection:

face, if product is handled such that it could be splashed into eyes.

Provide an eyewash station in the area.

Skin Protection: Impervious gloves (viton, nitrile) should be worn at all times when

> handling this material. In confined spaces or where the risk of skin exposure is much higher, impervious clothing should be worn. Safety

showers should be available for emergency use.

Respiratory Protection: If exposure exceeds occupational exposure limits, use an appropriate

NIOSH-approved respirator. Use a NIOSH-approved chemical cartridge respirator with organic vapour cartridges or use a NIOSH-approved supplied-air respirator. For high airborne concentrations, use a NIOSH-

approved supplied-air respirator, either self-contained or airline

breathing apparatus, operated in positive pressure mode.

9. PHYSICAL DATA

Physical State: Liquid Appearance: Bright Clear

Odour: Typical Gasoline Odour

Odour Threshold: Not available Freezing/Pour Point: <-51 degrees C **Boiling Point:** 60 - 260 degrees C

Density: 750 - 801 kg/m3 @ 15 degrees C

Vapour Density (Air = 1): Not available

Vapour Pressure: >42 mm Hg @ 38 degrees C

Specific Gravity (Water = 1):

pH: Not applicable

Flash Point: Method Tag Closed Cup = -23 - 1 degrees C

Lower Explosion Limit: 1.4 % (vol.) Upper Explosion Limit: 7.6 % (vol.)

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Autoignition Temperature: Not available
Viscosity: Not available
Evaporation Rate (n-BuAc = 1): Not available
Partition Coefficient (Kow): Not available
Water Solubility: Insoluble

Other Solvents: Hydrocarbon Solvents

10. STABILITY AND REACTIVITY

Chemically Stable: Yes
Hazardous Polymerization: No
Sensitive to Mechanical Impact: No
Sensitive to Static Discharge: Yes

Hazardous Decomposition Products: Thermal decomposition products are highly dependent on

combustion conditions.

Incompatible Materials: Avoid contact with strong oxidizing agents and acids.

Conditions of Reactivity: Avoid excessive heat, open flames and all ignition

sources.

11. TOXICOLOGICAL INFORMATION

Ingredient (or Product if not specified) Toxicological Data Naphtha (Petroleum), Full-range Reformed LD50 Oral Rat >28 mL/kg

Benzene LD50 Oral Rat = 930 - 5600 mg/kg

LC50 Inhaiation Rat = 13700 ppm for 4 hours

Routes of Exposure: Exposure may occur via inhalation, ingestion, skin absorption and skin or

eye contact.

Irritancy: This product is expected to be irritating to skin but is not predicted to be a

skin sensitizer.

Chronic Effects: Prolonged and repeated contact with skin can cause defatting and drying

of the skin resulting in skin irritation and dermatitis. Prolonged exposure to high vapour concentration can cause headache, dizziness, nausea, blurred vision and central nervous system depression. Prolonged and repeated exposure may cause serious injury to blood forming organs,

resulting in anemia and similar conditions.

Pre-existing Conditions: Pre-existing eye, skin and respiratory disorders may be aggravated by

exposure to this product.

Carcinogenicity and

Mutagenicity:

This product contains benzene. Epidemiological studies indicate that long term inhalation of benzene vapour can cause leukaemia in man. Benzene

has also produced chromosomal aberrations in peripheral blood

lymphocytes. Carcinogenic hazard.

12. ECOLOGICAL INFORMATION

SHELL JET B 141-012

Revision Number: 5

Environmental Effects Do not allow product or runoff from fire control to enter storm or sanitary

sewers, lakes, rivers, streams, or public waterways. Block off drains and ditches. Provincial regulations require and federal regulations may require that environmental and/or other agencies be notified of a spill incident. Spill area must be cleaned and restored to original condition or to the satisfaction of authorities. May be harmful to aguatic life. May

cause physical fouling of aquatic organisms.

Biodegradability Not readily biodegradable. Potential for bioaccumulation.

13. DISPOSAL CONSIDERATIONS

Waste management priorities (depending on volumes and concentration of waste) are: 1. recycle (reprocess), 2. energy recovery (cement kilns, thermal power generation), 3. incineration, 4. disposal at a licenced waste disposal facility. Do not attempt to combust waste on-site. Incinerate at a licenced waste disposal site with approval of environmental authority.

14. TRANSPORTATION INFORMATION

Canadian Road and Rail Shipping Classification:

UN/NA Number UN1863

Proper Shipping Name FUEL, AVIATION, TURBINE ENGINE

Hazard Class Class 3 Flammable Liquid

Packing Group PG II

Shipping Description FUEL, AVIATION, TURBINE ENGINE Class 3 UN1863 PG II

15. REGULATORY INFORMATION

This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all the information required by the Controlled Products Regulations.

WHMIS Class: Class B2 Flammable Liquid

Class D2B Other Toxic Effects - Skin Irritant
Class D2A Other Toxic Effects - Carcinogen

DSL/NDSL Status: This product, or all components, are listed on the Domestic Substances

List, as required under the Canadian Environmental Protection Act.

Other Regulatory Status: No Canadian federal standards.

16. ADDITIONAL INFORMATION

LABEL STATEMENTS

Hazard Statement : Flammable Liquid.

Irritating to skin.
Contains Benzene.
May cause cancer.

SHELL JET B 141-012

Revision Number: 5

Handling Statement: Eliminate all ignition sources.

Wear suitable gloves and eye protection.

Bond and ground transfer containers and equipment to avoid static

accumulation.

Avoid prolonged exposure to vapours.

Empty containers are hazardous, may contain flammable / explosive dusts,

liquid residue or vapours. Keep away from sparks and open flames.

First Aid Statement: Wash contaminated skin with soap and water.

Flush eyes with water.

If overcome by vapours remove to fresh air.

Do not induce vomiting.
Obtain medical attention.

Revisions: This MSDS has been reissued in the ANSI Z400.1 standard format.



Material Safety Data Sheet

WHMIS (Pictograms)	WHMIS (Classification)	Protective Clothing	TDG (pictograms)	
\bigcirc	A, B-1	0		

Product Nam	PROPANE	
Synonym	Propane HD-5. Propane commercial, Dimethylmethane, Propyl hydride, Liquefied Petrole	oum Gas (LPG), Alkane, C3H8
Supplier	ICG PROPANE Sulto 200, 19433 96th Avenue Surrey, BC V4N 4C4	In case of amerocney
Material Use	Propane is used as a fuel gas, refrigerant and as a raw material for organic synthesis. The grade determines the propane content. It is supplied as pressurized liquid in tanks and cylinders.	1-800-424-8807

				Exposure L	mits (ACGIH)	
	Name	CAS#	% (V∧)	TLV-TWA (8 h)	STEL	CEILING
Propane		74-98-6	>90	4508 mg / m3. (2500 PPM - 1898 Occupational Exposure Limits)	Not applicable	Not applicable
Propylene */** Butane Ethane Ethyl mercaplan		115-07-1 108-97-8 74-84-0 75-08-1	<5 <3 0-5 <50 ppm	Simple asphyxiant 800 ppm Simple asphyxiant 0.500m	Not applicable Not applicable Not applicable Not applicable	Not applicable Not applicable Not applicable Not applicable
*Propane commercial con **Propylene may not be p	, , , ,					, and application
Supplier Recommendation	Recommends a maximum exposure level of 2500 ppm (4508 mg/m3) for 8 hours time weighted average when handling propane based on 1996 ACGIH notice of intended change for propane, Consult local authorities for acceptable exposure limits.					
Other Exposure Limits	Consult local, provincial or temtory authorities for acceptable exposure limits.					

Section 3. Hazards Identification				
Potential	The health effects caused by exposure to propane are much less serious than it's fire and explosion risk. Propane is essentially			
Health	nontoxic in concentrations less than the lower explosive limit, but at very high concentrations it is a simple asphyxiant and			
Effects	displaces oxygen from the breathing atmosphere. Lack of oxygen may cause dizziness, headaches, diminished awareness, faulty judgement, increasing fatigue, impalred muscular coordination progressing to convulsions, come and death. A person working around propane in an enclosed space or in close proximity to a propane source (filling cylinders, purging lines and lighting / adjusting pilot lights, etc.) who feels "light-headed", "dizzy", "drunken" or a little intoxicated should realize this effect may be due to a dangerously high level of propane vapours (in the explosive range) and go immediately into fresh air. Direct contact with escaping gas or liquefied gas can result in freezing burns or frost bite to skin and eyes. For more information, refer to Section 11.			

Section 4. First Ald Measures				
Eye Contact	If the eye ilssue is frozen, seek medical attention immediately; if tissue is not frozen, immediately and thoroughly flush the eyes with running water for at least 15 minutes, keeping eyelids open, if irritation, pain, swelling, or crying has occurred, get medical attention.			
Skin Contact	Exposure to rapidly expanding gas or vaporizing flouid may cause frostbite (cold burns). If frostbite has occurred, do not rub the affected areas or flush them with water, but thaw frosted parts by soaking in water. In order to prevent further tissue damage, do not attempt to remove frozen clothing from frostbitten areas. If frostbite has not occurred, immediately and thoroughly wash contaminated skin with soap and water.			
Inhalation	Evacuate the victim to fresh air at once. If the victim is not breathing, perform mouth-to-mouth resuscitation, Administer oxygen if available. Keep the victim warm and at rest, Seek medical attention as soon as possible.			
Ingestion	Since the product is a gas and that it is mostly probable that it will be inhaled more than ingested, please consider to look first at the preventive measures in case of inhalation.			
Note to Physician	Monitor for respiratory distress. If cough or difficulty in breathing develops, evaluate for respiratory tract imitation, bronchitis, or pneumonitis. Monitor blood gases to assure adequate ventilation, if vital signs become abnormal or symptoms develop obtain a chest x-ray.			

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Available in French

Personal Protection	•
Eyes	Face shield, safety glasses or chemical splash goggles in case of splashing,
Body	Wear appropriate loose clothing with closed neck and long sleeves to prevent the skin from becoming frozen from contact with the liquid or from contact with vessels containing the liquid.
Respiratory	When exposure is likely to exceed recommended exposure limit (see section #2), use NIOSH approved respirator. Respirator should be selected based on the form and concentration of containment in air (refer to NIOSH Pocket Guide for Chemical Hazard for respirator selection). In order to determine the concentration of the containment, air sampling is recommended and should be performed by a health and safety specialist (as per the NIOSH Manual of analytical methods for method of measurement). If air sampling is not practical and concentration is unknown, use SCBA.
Hands	Wear insulated gloves to prevent frostbite.
Feet	Safety books or shoes.

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Section	9. Physical	and Chemical Properties	,	
Physical State & Appearance	Gas at room to pressure.	emperature; liquid when stored under	Viscosity	Not applicable.
Colour	Colourless.		Pour Point	Not applicable,
Propane sold has an odour mercaptan, w or "skunk".		in natural state at any concentration, rivel purposes under pressure usually added to it. This odourant is usually a hinas an odour similar to "rotten eggs" odourant level is such that it is the Lower Exposure Limit (LEL) of the	Softening Point	Not applicable
		ies have shown that not all persons are skunky smell and may not be able to ng device!		*
Odour Threshold	overexposure to	an adequate warning to prevent or propane. Prolonged exposure to cause olfactory desensitization.	Dropping Point	Not applicable.
Boiling Point	-42°C (-44°F).	Penetration	Not applicable.
Density	0.51 Kg/L @)15°C (Water = 1).	Oil/Water Dist. Coa	ff Log Kow: 2.36; mobile.
Vapour Densil	y 1 56 @ 0°C	(32°F), 1.8 @ 20°C (68°F) Air = 1.	Ionicity (in water)	Not applicable,
Vapour Pressu	apour Pressure <10763 mmHg @ 100°F (<1435 kPa @ 38°C).		Dispersion Properti	
Volatility	Volatile		Solubility	62ppm in water at 25°C (77°F), slightly soluble acetone. Soluble in benzene, ether, alcoho chloroform.
Section	10. Stability	and Reactivity		
Corrosivity	Non corresive			
Stability	The product is st conditions.	able under normal handling and storage	Hazardous Wi Polymerization	If not occur under normal conditions.
incompatible Substances/ Conditions to A	5: 10	lth oxidizing agents (percxides, chlorine).		eleases of Co., smoke and irritating fumes who eated to decomposition
Section	11. Toxicolo	gical Information		
Roules of Entr	У	Inhalation, skin contact and eye contact.		
Acute Lethality	,	Simple asphyxiant, LC50 (Inhalation/hum few minutes at 100,000 ppm (10%).	nan): no effect for 10	.000 ppm (1%) break exposure; slight dizziness in
Dermal Route Inhalation Ro Oral Route: Eye Irritation/	ule: Inflammation.	Low dermal penetration. Skin imitation human volunteers. Subchronic inhalation studies in monker No studies were found. None.		of organs toxicity or abnormalities.
Respiratory T Mutagenic: Reproductive Teratogenicity Carcinogenic	in Sensitization: No studies were found. Inspiratory Tract Sensitization: No studies were found, Integenic: Not mutagenic in the Salmonella typhimurium/microsome assay (Ames test). No studies were found. Interpretation of the Salmonella typhimurium/microsome assay (Ames test). No studies were found. Interpretation of the Salmonella typhimurium/microsome assay (Ames test). No studies were found. Interpretation of the Salmonella typhimurium/microsome assay (Ames test). No studies were found. Simple asphyxiant.			say (Ames test).
Carcinogenic	inogenicity (IRC) No studies were found. inogenicity (NTP) No studies were found inogenicity (IRIS) No studies were found inogenicity (OSHA) No studies were found.			

JUN-09-2000 09:41 ILG PRUPHNE INC. PROPANE Other Considerations Acts as a simple asphyxiant - inert gas or vapour. The narcotic or intoxicated effect of a simple asphyxiant may impair a person's judgement, but it is temporary and will rapidly disappear in fresh air. Persons with anemia or other conditions of reduced oxygen-carrying capacity may be more sensitive. Propane producers and distributors may, from time to time, add small amounts of methanol to the propane to overcome water and freezing problems. Methanol may accumulate in liquid residues in propane piping and storage vessels. Please refer to a methanol Material Safety Data Sheet (MSDS) for further details concerning methanol. Section 12. Ecological Information Volatilizes and disperses rapidly. Volatilation is Persistence/ Environmental Propene is readily biodegraded by soll bacteria Fate expected to be the dominant fate process. Bloaccumulation (Microbacterium vaccae) The degradation of propane Potential is similar to the degradation of fatty acids. Products of Not available. BODS & COD Not available. Biodegradation Additional Remarks Henry's Law constants for propage has been calculated to be 7.07x10-1 atm-m/mole @ 25°C. These mean that propage may rapidly volatilize from water and moist soil to the atmosphere. The estimated half-life for evaporation of propane from a model river (1m deep flowing 1m/s with a wind speed of 3 m/s) and a model pond are 1.9 hrs. and 2.3 hrs., respectively. Section 13. Disposal Considerations Preferred waste management priorities are: (1) incineration with energy recovery; (2) evaporation; (3) disposal at licensed Waste Disposal waste disposal facility. Ensure that disposal or reprocessing is in compliance with government requirements and local disposal regulations. Section 14. Transport Information 102 Add "SPECIAL COMMODITY" to document if in Shipping name: Propane or Liquefied Special TDG Classification Petroleum Gas: UN1978 or UN 1075, Class 2.1; Provisions for car load, container load by rail. Acceptable modes of transportation, air (cargo only), rail, road and water, Label required: Flammable Gas. Transport Not acceptable for transport by passenger aircraft. Section 15. Regulatory Information All components of this formulation are listed in the Domestic Substances List (DSL-Canadian) and in the Toxic Substances Other Regulations Control Act Inventory (TSCA-U.S.). This product is not known to contain any of the carcinogens required to be listed under OSHA hazard communication standard, 29 CFR 1910.1200 (U.S.). Not listed in EPCRA or SARA Title III, Section 313, Toxic Chemicals (40 CFR 355), Not listed in CERCLA (40 CFR 302.40), Please note that the chemical identity of some or all of the ingredients that may be listed herein is confidential business information and is being withheld as permitted by 29 CFR 1910.1200 and various State Right to Know Laws. 2- Risk of explosion by shock, friction, fire or other sources of ignition, 13- Extremely flammable liquefied gas, 16- Explosive DSD/DPD (Europa) when mixed with oxidizing substances, 20/21- Harmful by inhalation and in contact with skin, 35- Causes severe burns. DOT (U.S.A.) (Pictograms) Rating Health Hazard NFPA (U.S.A.) HMIS (U.S.A.) 0 Insignificant DIASEH GIL Fire Hazard 4 1 Slight 2 Moderate Reactivity 0 3 High Personal Protection 4 Extreme Section 16 Other Information

References Available upon request.		
Glossary	DSD/DPD - Dangerous Substance or Dangerous Preparations Directives (Europe)	
ACGIH - American Conference of Governmental Industrial Hygienists	DSL - Domestic Substance List	
ASTM - American Society for Testing and Materials	EEC/EU - European Economic Community/European Union	
BOD5 - Biological Oxygen Demand in 5 days	EINECS - European Inventory of Existing Commercial Chemical	
CAN/CGA B149.2 - Propage Installation Code	Substances	
CAS - Chemical Abstract Services	EPCRA - Emergency Planning and Community Right-to-Know Act	
CEPA - Canadian Environmental Protection Act	FDA - Food and Drug Administration	
CERCLA - Comprehensive Environmental Response, Compensation	FIFRA - Federal Insecticide, Fungicide, and Rodenticide Act	
and Liability Act	HCS - Hazardous Communication System	
CFR - Code of Federal Regulations	HMIS - Hazardous Material Information System	
CHIP - Chemical Hazard Information and Packaging Approved Supply	IARC - International Agency for Research on Cancer	
List	IRIS - Integraled Risk Information System	
COD - Chemicai Oxygen Demand	LO50/LC50 - Lethal Dose/Concentration kill 50%	
CPR - Controlled Products Regulation	LDLo/LCLo - Lowest Published Lethal Dose/Concentration	
DOT - Department of Transportation	NAERG'96 - North American Emergency Response Guide Book (1996	
DSCL - Dangerous Substances Classification and Labeling (Europe)	NFPA - National Fire Prevention Association	

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NIOSH - National Institute for Occupational Safety & Health NPRI - National Pollutant Release Inventory NTP - National Toxicology Program OSHA - Occupational Safety & Health Administration PEL - Permissible Exposure Limit RCRA - Resource Conservation and Recovery Act SARA - Superfund Amendments and Reorganization Act SD - Single Dose STEL - Short Term Exposure Limit (15 minutes)	TDG - Transportation Dangerous Goods (Canada) TDLo/TCLo - Lowest Published Toxic Dose/Concentration TLV-TWA - Threshold Limit Value - Time Weighted Average TSCA - Toxic Substances Control Act USEPA - United States Environmental Protection Agency USP - United States Pharmacopoeia WHMIS- Workplace Hazardous Material Information System
Information Contact ICG Propose Inc. 1-800-424-8807	Prepared by: Williams 99/12/09 Data entry by: Data Business Forms

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