Material Safety Data Sheet / Fiche signalétique

WESTCOAST DRILLING SUPPLIES LTD.

8069 River Way, Delta, British Columbia,

Canada V4G 1L3

Ph. (604) 940-6050 Fax (604) 940-6080

EMERGENCY 1-800-665-6645

SECTION I: IDENTIFICATION OF PRODUCT

PRODUCT NAME:

CHEMICAL FAMILY:

PELADOW®

PRODUCT USE:

Oil Well Fluid Additive Calcium Chloride (>90%)

WORK PLACE HAZARDOUS MATERIALS INFORMATION SYSTEM (WHMIS)

WHMIS CLASSIFICATION:

Class D-2(B)

WORK PLACE HAZARD:

Skin and Eye Irritant

TRANSPORTATION OF DANGEROUS GOODS (TDGR)

SHIPPING NAME:

Not Regulated Under TDG

TDG CLASSIFICATION:

Not applicable

PACKAGE GROUP:

Not applicable

PRODUCT IDENTIFICATION NUMBER (PIN):

Not applicable

SECTION II: HAZARDOUS INGREDIENTS

INGREDIENT PERCENTAGE
Calcium Chloride 91 - 92%

CAS NUMBER LD50 (oral rat)

Calcium Chloride
Sodium Chloride
Potassium Chloride

10043-52-4 7647-14-5 900-2100 mg / kg (100%) Not available Not available

Potassium Chloride Strontium Chloride 2-3% 1%

1 - 2%

7447-40-7 10476-85-4

Not available

SECTION III: TOXICOLOGICAL PROPERTIES

ROUTE OF ENTRY:

[XX] Skin, [XX] Eye Contact, [XX] Inhalation, [XX] Ingestion

EYE CONTACT:

Dusts may cause moderate to severe eye irritation. with corneal

injury that may be slow to heal. When dissolving, the heat produced

may cause more intense effects as well as thermal burns.

SKIN CONTACT:

Short single exposure not likely to cause significant skin irritation. Prolonged or repeated exposure may cause skin irritation, even a burn. May cause more severe response if skin is damp and/or abraded, or if material is confined to skin. When dissolving, the heat

produced may cause more intense effects as well as thermal burns. DOT classification: Noncorrosive.

SKIN ABSORPTION:

A single prolonged exposure is not likely to result in the material being absorbed through the skin i harmful amounts. The LD₅₀ for

skin absorption in rabbits is >5000 mg / kg.

INGESTION:

Single dose oral toxicity is low. Swallowing solids may cause

gastro-intestinal irritation or ulceration.

INHALATION:

Vapours are unlikely due to physical properties. Dust may cause

irritation of upper respiratory tract.

SECTION IV: FIRST AID MEASURES

SKIN CONTACT: Immediately wash skin with lots of soap and water. Remove contaminated clothing and shoes; wash before reuses. Get medical attention if irritation persists after washing.

EYE CONTACT: Get immediate medical attention. Immediately flush eyes with lots of running water for 15 minutes, lifting the upper and lower eyelids occasionally.



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INGESTION: If conscious, immediately induce vomiting. Get immediate medical attention. Do not give anything by mouth to an unconscious or convulsing person.

INHALATION: remove to fresh air. Give artificial respiration if not breathing. Get immediate medical attention.

NOTE TO PHYSICIAN: If burn is present, treat as any thermal burn, after decontamination, No specific antidote. Support-

ive care. Treatment based on judgement of the physician in response to reactions of patient.

SECTION V: PHYSICAL DATA

APPEARANCE AND ODOR:

White to off-white powder; odorless.

DENSITY (SPECIFIC GRAVITY):

2.2 g/cc

BOILING POINT:

198° C

MELTING POINT:

Approximately 230° C

WATER SOLUBILITY:

Soluble pH 7-10

% VOLATILE BY VOLUME:

Not applicable

EVAPORATION RATE:

Not applicable

VAPOR PRESSURE: (mm Hg)

0.009 mm Hg 20° C

VAPOR DENSITY: (Air = 1)

Not applicable

SECTION VI: FIRE AND EXPLOSION HAZARD DATA

FLASH POINT:

Not applicable

FLAMMABLE LIMIT:

Not applicable

EXTINGUISHING MEDIA:

This material is not combustible

SPECIAL FIRE FIGHTING PROCEDURES:

Self-contained respirators required for fire-fighting personnel

UNUSUAL FIRE AND EXPLOSION HAZARDS:

None

SECTION VII: REACTIVITY DATA

STABLE [XXX] INSTABLE []

INCOMPATIBILITY (CONDITIONS TO AVOID):

(Specific Materials to Avoid) Calcium Chloride will: corrode most metals exposed to air, attack aluminium (and its alloys) and yellow brass; react with sulphuric acid to form hydrogen chloride which is corrosive, irritating, and reactive; give an exothermic reaction with water-reactive materials such as sodium; result in a runaway polym erization reaction with methyl vinyl ether (Bretherick, 1979); and, in solution form react with zinc (galvanizing) to yield hydrogen gas which is explosive (Ibid.). (Bretherick, L. 1979, Handbook of

Reactive Chemical Hazards, 2nd Ed.)

HAZARDOUS DECOMPOSITION PRODUCTS:

HAZARDOUS POLYMERIZATION:

Not Applicable

Will not occur [XX] May occur []

SECTION VIII: PREVENTATIVE MEASURES

RESPIRATORY PROTECTION:

Approved dust respirator mask

VENTILATION:

Local mechanical exhaust

PROTECTIVE GLOVES:

Rubber gloves.

EYE PROTECTION:

Chemical goggles.

OTHER PROTECTIVE EQUIPMENT:

An eyewash and safety shower should be nearby and ready for use.



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PRECAUTIONS TO BE TAKEN IN HANDLING AND STORAGE:

Avoid eye and prolonged skin contact, Always use cool water (temperature less than 27° C), when dissolving calcium chloride. Heat developed by solutions is very high during dissolving and mixing. When exposed to atmosphere, calcium chloride will pick up water and form a solution. Leather clothing and shoes will be damaged by calcium chloride.

STEPS TO BE TAKEN IN CASE OF SPILL OR LEAK:

Losses incidental to correct applications of this product in its intended uses are not expected to be harmful to the environment. Wear appropriate safety apparel during clean-up. Avoid entry of large amount of product into sewers, natural waters, and drinking water sources. Due to possible harmful effects. Avoid contact with vegetation, animals and fish life. Recover quickly into suitable, dry sealable containers if reusing. Small quantities may be flushed away with plenty of water. Walking surfaces may remain wet longer due to moisture being held by spilled product - avoid by thoroughly water washing surfaces.

WASTE DISPOSAL METHOD:

Dispose of contaminated product and materials used in cleaning up spills or leaks in a manner approved for this material. Consult appropriate regulatory agencies to ascertain proper disposal procedures

SECTION IX: PREPARATION

The information contained herein is given in good faith, but no warranty, expressed or implied is made.

DATE ISSUED: December 15, 1994

BY: Product Safety Committee

DATE REVISED: April 1, 2000

Review date;

Authorize