

# ***Kimmirut Wetland Planning Study***

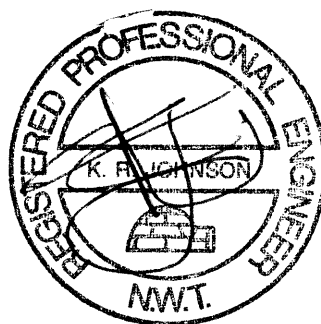
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## Introduction

This report presents an estimate of wetland water quality improvement performance based on the operations plan of September 28, 2007 (Trow, 2007). The suggested configuration of the Kimmirut system includes retaining wastewater in winter in the existing (modified) lagoon and a second lagoon at the base of the downstream gulch. The area between the lagoons (gulch) will be used to retain the winter wastewater icefield. These new treatment and storage facilities are needed to contain the entire winter discharges from the town. The end of gulch lagoon discharges to an extensive wetland, which will provide further treatment.

The total annual design flow from Kimmirut is expected to be approximately 33,000 m<sup>3</sup>. Winter wastewater storage, summer wastewater flows, and runoff will be discharged from the expanded lagoon system over a 107-day warm window of opportunity. The unfrozen period is assumed to span mid-June to late September, or three and a half months, during which the water temperatures are estimated to be 5°C. Flows from the gulch currently flow down a steep incline (>40%) to a large wetland at lower elevation, and it is envisioned that future flows will follow this same path. That wetland currently drains from north to south, away from Lake Fundo, toward Lake Tulsit (Dillon, 2001). The large lower wetland has been identified as muskeg, with dwarf birch, willow shrubs and grasses; but has also been termed a river valley (Dillon, 2001). These descriptions are consistent with the observation that the lower wetland, of more than 15 hectares, is a channelized area of multiple streams embedded in the wetland (**Figure 1**). The wetland slopes gently to the south, at about 0.5%. Because of channelization, it is estimated that only about 20% of the wetland area, or 3 ha, would be in the flow path of the new wastewater lagoon discharges. The assumption is that the lagoon flow will use 20% of the wetland area, and effectively stay separate from the rest of the wetland and the runoff flow going through it. The real situation would probably be somewhere between this assumed segregated flow and completely mixed flow, but here the effects of dilution are not considered.

As an estimate of treatment potential, it is assumed that lagoon treatment will occur in the unfrozen lagoons, and that the combined meltwater and warm weather flows will have BOD and TSS of approximately 230 and 245 mg/L respectively. The discharge season is the three and a half month unfrozen period, during which the water temperatures are estimated to be 5°C, with a range of only from 2 – 8 °C. That period will see flows from the wastewater glacier and lagoons, which are controlled as indicated in Trow (2007), accompanied by melting and seepage phenomena. Stored water is estimated to be gone by the end of the warm season. The warm season flow to the wetlands will average approximately 761 m<sup>3</sup>/d, or about eight (8) times the daily rate of generation.

## Wetland Forecasting Procedure

Three principal themes have been prevalent in the history of constructed treatment wetland design: consideration of the pollutant and hydraulic loadings, first order removal models, and regression equations. The strategy advocated here is to utilize the existing data sets to maximum advantage, while retaining as much simplicity as possible. The strategy used here attempts to salvage the good features of past work, and to improve upon past methods as much as data can support. The strategy is not simpler than previous methods, because it includes more pieces; that is, it is an integrated rational approach to forecasting. The calculations are implemented via spreadsheets on a desktop computer, because of the level of detail in the procedure.



**Figure 1. Location of Wetland**

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The key features of the forecasting procedure are:

Basis:

1. Set inlet concentrations
2. Set inflow and seepage
3. Determine rain, evapotranspiration (ET), temperature
4. Select wetland area

Parameter Selection and Calculation:

5. Select rate constants and seasonality
6. Select detention time distribution (DTD) (hydraulic) efficiency = P value (Select compartmentalization).

Constraint Checks:

7. Set estimated growth cycle
8. Check biogeochemical cycles for consistency
9. Check constraints (hydraulic and chemical)
10. Check loading graph for intersystem comparison.

### Water Budget

The annual water budget forms the basis for a first approach to understanding pollutant reductions and the area required. Under the assumption of a constant water level, the flows out of the wetland may be computed from the inflow and the meteorological data. Some infiltration is assumed here, if leakage is present in the wetland. The relevant equations are:

$$Q_{out} = Q_{in} + A \cdot (P - ET - I) \quad (1)$$

or  $q_{out} = q_{in} + (P - ET - I) \quad (2)$

where

A	=	wetland area, m <sup>2</sup>
ET	=	evapotranspiration rate, m/d
I	=	infiltration rate, m/d
P	=	precipitation rate, m/d
q	=	hydraulic loading rate, m/d
Q	=	flow rate, m <sup>3</sup> /d

The inlet hydraulic loading will be increased by rainfall (ca. 1.6 mm/d in the discharge season), and decreased by evapotranspiration (ca 3.0 mm/d). These amounts are important if the wetland is to have a very low hydraulic loading, or correspondingly, a long detention time. The ratio  $\alpha = (P-ET)/q_i$  is the atmospheric augmentation.

The pollutant mass balances will be conducted on a cells-in-series basis. Results of the overall water mass balance are apportioned to the cells according the chosen number of tanks in series (PTIS), which is presumed to be three, the average of tracer testing results of a large number of wetlands. For the first unit in the series:

$$Q_1 = Q_{in} + A_1(P - ET - I) \quad (3)$$

where:  $A_1$  = area of tank number 1,  $m^2$   
 $Q_1$  = flow rate out of tank 1,  $m^3/d$

Flows are thus computed sequentially, from inlet to outlet, for the number of tanks chosen (PTIS = 3).

### Flow Rates

The flow from the community is expected to be 91  $m^3/d$ , for a total of 33,356  $m^3/yr$ . This flow is augmented by 48,106  $m^3/yr$  of precipitation accumulated in the drainage area of the lagoons and gulch. The total water leaving the area of the two lagoons is thus 81,462  $m^3/yr$ , which is discharged over the period of 107 days. Thus the average flow to the wetlands is 761  $m^3/d$  during discharge.

Seepage is expected to be negligible, because of the hard rock base of the region, presumably including the wetland.

### Pollutant Mass Balances

Water passes through P tanks in series, and loses contaminant in each. For the case of no water losses or gains, the steady flow contaminant mass balance for the  $j^{th}$  tank is:

$$(QC_{j-1} - QC_j) = kA(C_j - C^*) \quad (4)$$

where  $C_j$  = concentration in and leaving tank "j",  $gm/m^3$   
 $C^*$  = background concentration,  $gm/m^3$   
 $k$  = removal rate constant,  $m/d$   
(or with unit conversion,  $m/yr$ )

For the entire sequence of tanks, these mass balances combine to:

$$\frac{(C - C^*)}{(C_0 - C^*)} = \left(1 + \frac{k\tau}{Ph}\right)^{-P} \quad (5)$$

where  $C_o$  = concentration entering the wetland, gm/m<sup>3</sup>  
 $h$  = water depth, m  
 $P$  = number of tanks in series, PTIS  
 $\tau$  = nominal detention time, d

Note that there are two reaction parameters in this model: the rate constant "k" and the hydraulic parameter "P". However, wetlands typically have inputs of rain, and outputs of evaporation and infiltration. Therefore, the flows are not spatially uniform, and a tank to tank calculation must be invoked. Also note that for very large P-values, this model becomes the exponential, plug flow model of Kadlec and Knight (1996).

The tanks in series model is then carried forward via a sequential calculation of pollutant concentrations for each "tank" in the chosen hydraulic model.

A first order areal model with rate constant k is selected, with a wetland background concentration C\* as necessary. The pollutant mass balance for the first of the wetland segments, designated by subscript "1", for steady state, non-uniform flow is:

$$Q_1 C_1 = Q_{in} C_{in} - I \cdot A_1 C_1 - \alpha E T A_1 C_1 - k \cdot A_1 \cdot (C_1 - C^*) \quad (6)$$

In this simple version, rainfall has been assumed to have a zero pollutant concentration, but it is easy to add an atmospheric input of the pollutant if it exists. Infiltration is assumed to occur at the outlet concentration. Transpiration flow of the contaminant has been included. Combining (3) with (6) gives the concentration exiting hypothetical segment number one:

$$C_1 = \frac{Q_{in} C_{in} + k \cdot A_1 \cdot C^*}{Q_1 + (\alpha E T) \cdot A_1 + I \cdot A_1 + k \cdot A_1} \quad (7)$$

or

$$C_1 = \frac{q_{in} C_{in} + k \cdot C^*}{q_1 + (\alpha E T) + I + k} \quad (8)$$

Note that the hydraulic loading rates in (8) are the tank loading rates, not the system loadings. This computation is then repeated sequentially for the remaining segments, in each case using the outlet concentrations and flows from the preceding unit. The wetland outlet concentration is that exiting from the final hypothetical segment.

The additional input data requirements for the pollutant mass balances are:

1. Input concentration ( $C_{in}$ );
2. Background concentration ( $C^*$ );
3. Transpiration fraction ( $\beta$ );
4. Rate coefficient (k).



### *Inlet Concentrations*

The wastewater from the community has high strength, due to the restricted use of water. However, water reaching the wetlands will have that strength reduced, due to (1) lagoon treatment, (2) cascade oxygenation, and (3) mixing with runoff water. There will be about forty (40%) percent dilution with precipitation water. The inlet concentrations in **Table 1** are estimated to be approximately half of the raw water strength.

### *Background Concentrations*

Wetland systems are dominated by plants (autotrophs), which act as primary producers of biomass. However, wetlands also include communities of microbes (heterotrophs) and higher animals, which act as grazers and reduce plant biomass. Most wetlands support more producers than consumers, resulting in a net surplus of plant biomass. This excess material is typically buried as peat or exported from the wetland. The net export results in an internal release of particulate and dissolved biomass to the water column, which is measured as non-zero levels of BOD, TSS, TN, and TP. These wetland background concentrations are typically denoted by the term  $C^*$ . Enriched wetland ecosystems (such as those treating wastewater) are likely to produce higher background concentrations. The background concentrations in **Table 1** are estimated from the literature.

### *Transpiration*

The fraction of water loss due to transpiration is assumed to be twenty (20%) percent. This flow draws water into the root zone for enhanced treatment.

### **Contaminant Removal Rate Coefficients**

The rate coefficients used to calculate estimated removals vary from substance to substance. The mechanisms and chosen values are discussed in this section.

### *Suspended Solids*

A major function performed by wetland ecosystems is the removal of suspended sediments from water moving through the wetland. These removals are the end result of a complicated set of internal processes, including the production of transportable solids by the wetland biota.

Low water velocities, coupled with the presence of litter on the wetland bottom, promote fallout and trapping of solid materials. This transfer of suspended solids from the water to the wetland sediment bed has important consequences both for the quality of the water and the properties and function of the wetland ecosystem.

A wetland processes sediments and suspended solids in a number of ways. After the suspended material reaches the wetland, it joins large amounts of internally-generated suspendable materials, and both are transported across the wetland. Sedimentation and trapping, and resuspension, occur en route, as does "generation" of suspended material by activities both above and below the water surface. Chemical reactions may cause the formation of solids that can deposit. Algal and plant debris may form at one location and deposit downgradient in the wetland. Thus it is clear that the TSS leaving a treatment wetland of moderate to long detention is more reflective of generation and resuspension than of unsettled incoming solids.

With P & ET, N TIS

T =	5.0	°C	
I =	0.00	cm/day	
P =	0.16	cm/day	
ET =	0.30	cm/day	
Trans =	0.06	cm/day	0.200 = T/ET
$\alpha$ =	-0.14	Augmentation	

$Q = 761 \text{ m}^3/\text{d} \quad 0.201 \text{ mgd}$

$l_{eff} = 0.22$  m/yr Effective Leakage  
 $\psi = 1.00$  Fraction  $NH_4$ -N Nitrified

 $C_i =$ 

TSS	BOD	TP	Org-N	NH <sub>4</sub> -N	NO <sub>x</sub> -N	TN	TKN	FC
245	230	5.00	20.00	40.00	0.00	60.00	60.00	10,000.000

62.2	58.4	1.3	5.1	10.2	0.0	15.2	15.2
------	------	-----	-----	------	-----	------	------

Ce =

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--	--	--	--	--	--	--	--	--

[illegible]

Cd =

[illegible] $C^*$  =

--	--	--	--	--	--	--	--	--

k20

20.0	20.0	0.002	1.50	0.00	0.00	1.50	50
------	------	-------	------	------	------	------	----

 $k =$ 

35	35	2	10	8	27	15	83
1,000	1,000	1,000	1,000	1,050	1,100	1,060	1,000

0.3

35	35	2	10	4	6	6	83
HRT =		11.6	days	Area =		322,780 ft <sup>2</sup>	
HLR in		2.54	cm/d			7 acres	

HLR in =	2.34	cm/d					7	acres
HLR out =	2.39	cm/d					3	ha
TSS	BOD	TP	Org-N	NH <sub>4</sub> -N	NO <sub>x</sub> -N	TN	TKN	FC

39.2	37.9	3.97	8.7	34.7	10.7	54.1	43.5	156.918
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9.38	9.07	0.95	2.09	8.32	2.55	12.95	10.40
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Accordingly, the estimated k-value for TSS has been chosen as 35 m/yr, with no effect of water temperature. In any case, the water temperature in the wetlands is expected to range only from 2 – 8°C, with a mean discharge season value of 5°C. This rate coefficient is sufficiently high to drive the wastewater TSS down close to the background value of 20 mg/L.

### **Biochemical Oxygen Demand**

Carbon compounds interact strongly with wetland ecosystems. The carbon cycle in wetlands is vigorous and typically provides carbon exports from the wetland to receiving ecosystems. Many internal wetland processes are fueled by carbon imports and by the carbon formed from decomposition processes.

Treatment wetlands frequently receive large external supplies of carbon in the added wastewater. Any of several measures of carbon content may be made, with Biochemical Oxygen Demand (BOD) being the most frequent in the treatment of municipal wastewater. Degradable carbon compounds are rapidly utilized in wetland carbon processes. At the same time, a variety of wetland decomposition processes produce available carbon. The balance between uptake and production provides the carbon exports. In general, the amounts of carbon cycled in the wetland are comparable to the quantities added in domestic wastewater. The wetland cycle of growth, death, and partial decomposition uses atmospheric carbon, and produces gases, dissolved organics, and solids. Decomposition involves the sugars, starches and low molecular weight celluloses in the dead plant material. Gaseous products include methane and regenerated carbon dioxide. A spectrum of soluble large organic molecules, collectively termed humic substances, are released into the water. The solid residual of plant decomposition is peat or organic sediment, which originated as celluloses and lignins in the plants. These wetland soil organics are broadly classified as fulvic material, humic material, and humin.

The P-k-C\* first order model can readily account for observations, for appropriate values of parameters. There are four (4) levels of inlet concentration to be considered: tertiary ( $0 < C_i < 30$  mg/L); secondary ( $30 < C_i < 100$  mg/L); primary ( $100 < C_i < 200$  mg/L); and "super" ( $C_i > 200$  mg/L). The Kimmirut situation is expected to be in the fourth group, with high incoming BOD. At high strength, the k-values for BOD are relatively high, with a median value of 189 m/yr for forty-three (43) wetland datasets. Further, temperature effects have been found to be minimal. For purposes of forecasting,  $k = 35$  m/yr has been selected, which is the 20<sup>th</sup> percentile of the distribution across wetlands (Kadlec and Wallace, 2008). This rate coefficient is sufficiently high to drive the wastewater BOD down close to the background value of 20 mg/L.

### **Nitrogen**

Nitrogen compounds are among the principal constituents of concern in wastewater because of their role in eutrophication, their effect on the oxygen content of receiving waters, and their toxicity to aquatic invertebrate and vertebrate species. These compounds also augment plant growth, which in turn stimulates the biogeochemical cycles of the wetland. The most important inorganic forms of nitrogen in wetlands treating municipal or domestic wastewater are ammonia ( $\text{NH}_4^+$ ) and oxidized nitrogen ( $\text{NO}_2^-$  and  $\text{NO}_3^-$ ). Nitrogen is also invariably present in wetlands in organic forms. Both dissolved and particulate forms may be present, but in most cases there is little particulate nitrogen in settled wetland surface waters.

## Organic Nitrogen

Wastewaters contain varying amounts of organic nitrogen, depending upon the source. Nitrogen in domestic sewage is comprised of about sixty (60%) percent ammonia, and forty (40%) percent organic nitrogen. Lagoon effluents may retain approximately the same proportions while reducing total nitrogen. For Kimmirut, it has been assumed that there are 20 mg/L of organic nitrogen that survive passage through the two lagoons and the two aeration cascades, compared to 40 mg/L of ammonia nitrogen.

Ammonification is the biological transformation of organic nitrogen to ammonia and is the first step in mineralization of organic nitrogen. This process occurs both aerobically and anaerobically, and releases ammonia from dead and decaying cells and tissues. Kinetically, ammonification proceeds more rapidly than nitrification, thus creating the potential for increasing ammonia concentrations along the flow-path of a wetland and requiring design for nitrogen removal to include both ammonification and the slower nitrification process. The ammonification process does not proceed to completion in wetlands, although the removal of ammonia can go to completion for long enough detention. There is an organic nitrogen background concentration which may consist of irreducible residuals, or be due to return fluxes of organic nitrogen from decomposing solids, which is typically 1.5 mg/L.

The median net period-of-record removal rate for 54 FWS systems receiving more than 5 mg/L of organic N is 97 g/m<sup>2</sup>•yr. The median k-value for organic nitrogen is 17.3 m/yr, but the range is wide. The 10<sup>th</sup> to 90<sup>th</sup> percentile range is 3.6 – 39.9 m/yr. For purposes of forecasting, k = 10 m/yr has been selected, which is the 25<sup>th</sup> percentile of the distribution across wetlands (Kadlec and Wallace, 2008). There appears to be little or no temperature dependence of organic nitrogen k-values.

## Ammonia Nitrogen

Ammonia is an intermediate in the sequential processing of nitrogen in treatment wetlands, which is produced by ammonification of organic nitrogen, and reduced by aerobic and possibly anaerobic ammonia oxidation processes. Because of toxicity of free ammonia in receiving aquatic ecosystems, this nitrogen species is often singled out for regulation. Free ammonia depends upon water temperature as well as total dissolved ammonia.

When the TN loading to the wetland is less than the growth requirements of the plants and algae by a considerable margin, the removal of TN is very likely to be mediated by the growth and decay of biomass. As a rough guideline, this situation occurs for TN loading less than approximately 120 gmN/m<sup>2</sup>•yr. In the Kimmirut situation, plant growth requirements are insignificant, and the processing of ammonia is likely to be entirely due to microbial functions.

The median net period-of-record removal rate for one hundred and eighteen (118) systems receiving more than 1 mg/L ammonia N is 127 g/m<sup>2</sup>•yr. The value C\* = 0.0 mg/L is used, and the remaining model parameter is the k-value. Calibration included ammonification (production) and nitrification (destruction), as well as return of organic nitrogen from the decomposition of biomass. The median annual rate constant was k = 14.7 m/yr. The 10<sup>th</sup> to 90<sup>th</sup> percentile range is 4.7 – 85.6 m/yr. For purposes of forecasting, k<sub>20</sub> = 8 m/yr has been selected, which is the 25<sup>th</sup> percentile of the distribution across wetlands (Kadlec and Wallace, 2008). There is a significant temperature dependence of ammonia k-values, with a theta value of 1.05. Therefore, at the mean water temperature of 5°C, the rate coefficient for ammonia is, k = 4 m/yr.

## Oxidized Nitrogen

Nitrate is potentially tied quite closely to the process of nitrification in wetlands that receive both ammonia and oxidized nitrogen, because incoming nitrate loads may be supplemented by produced nitrate. For wetlands which receive and reduce large amounts of organic and ammonia nitrogen, the inferred denitrification is much different from the net loss or gain of nitrate from inflow to outflow. Nitrate is entirely consumable in treatment wetlands. The value for  $C^*$  is zero, because no investigation has shown a lower limit to the reduction of nitrate.

Seventy-two (72) nitrate-dominated wetlands were calibrated for  $k$ . The median annual rate constant was  $k_{NN} = 26.5$  m/yr, while the average was  $k_{NN} = 30.0$  m/yr. The 10<sup>th</sup> to 90<sup>th</sup> percentile range is 9.6 – 54.4 m/yr (Kadlec and Wallace, 2008). For purposes of forecasting,  $k_{20} = 27$  m/yr has been selected, which is the 50<sup>th</sup> percentile of the distribution across wetlands (Kadlec and Wallace, 2008). There is a significant temperature dependence of nitrate  $k$ -values; thus even on an average annual basis, temperature or season may be an important determinant of the rate constant, and these factors are thus responsible for some of the intersystem variability in annual  $k$ -values. The median theta value for twenty (20) wetlands was 1.10 (Kadlec and Wallace, 2008). Therefore, at the mean water temperature of 5°C, the rate coefficient for nitrate is  $k = 6$  m/yr.

## Phosphorus

Treatment wetlands are capable of phosphorus (P) removal from wastewaters, on both short-term and long-term bases. Phosphorus is a nutrient required for plant growth, and is frequently a limiting factor for vegetative productivity. P-cycling in wetlands may be visualized as consisting of several compartments: water, plants, microbiota, litter, and soil. There are three principal categories of P removal processes in wetlands: sorption, utilization to build a bigger biomass compartment, and storage as newly created, refractory residuals (burial). The first two are transitory processes, and the corresponding capacities are used up. The third, accretion of new materials with structural phosphorus, is sustainable. There is a misconception that wetlands provide phosphorus removal only through sorption processes on existing soils. It is true that most soils do have sorptive capacity for phosphorus, but this storage is soon saturated under any increase in phosphorus loading. There are two direct effects of vegetation on phosphorus processing and removal in treatment wetlands:

1. The plant growth cycle seasonally stores and releases P, thus providing a "flywheel" effect for a P removal time series.
2. The creation of new, stable residuals which accrete in the wetland. These residuals contain phosphorus as part of their structure, and hence accretion represents a burial process for P.

Despite the apparent complexity of the several removal mechanisms, data analysis shows that relatively simple equations can describe the sustainable processes. Profiles of phosphorus concentration within wetlands typically show a decreasing trend that approaches the background concentration asymptote if the wetland is large enough. Global first order removal rates characterize such removal profiles, but do not incorporate any biotic features. Results across systems are given here, for measured P-values, or for the mean of those ( $P = 3.4$ ) when not measured. The value  $C^* = 0.002$  mg/L is used, and the remaining model parameter is the  $k$ -value, selected to fit the model. Across 282 wetlands, the median annual rate constant was  $k = 10.0$  m/yr. The 10<sup>th</sup> to 90<sup>th</sup> percentile range is 1.4 – 60 m/yr. (Kadlec and Wallace, 2008). For purposes of forecasting,  $k_{20} = 2$  m/yr has been selected, which is the 15<sup>th</sup> percentile of the distribution across wetlands (Kadlec and Wallace, 2008). This low value is selected because of the perceived low amount of biological activity in the Kimmirut wetland. There appears to be little or no temperature dependence of phosphorus  $k$ -values.





## Pathogens

Pathogens are present in untreated domestic wastewaters. Wetlands have been found to reduce pathogen populations with varying but significant degrees of effectiveness. Bacteria, protozoa, helminths and viruses typically do not survive longer than about thirty (30) days in freshwater environments, and about fifty (50) days in soil environments. Similar survivals might therefore be predicted for wetlands, but there are many site specific factors and processes which may materially increase or decrease survival. Ultraviolet radiation is a potent agent for killing bacteria. Most bacteria are food for nematodes, rotifers and protozoa. Among these, rotifers and flagellated and ciliated protozoa have been implicated as important contributors to the reduction of bacteria in treatment wetlands. A measurable proportion of wastewater microorganisms are found either associated with particulates, or as aggregates of many organisms, and are removed by particulate settling and trapping.

First-order models have been used to describe reduction of indicator bacterial populations in lagoons and wetlands. Data from twenty-eight (28) FWS systems, totaling forty-seven (47) wetland-years, was used to estimate fecal coliform k-values for a presumed  $C^* = 40/100\text{ml}$ . These systems were selected for having at least 1,000/100ml inlet FC, thus eliminating systems with low incoming FC, whether due to pretreatment disinfection or other factors. Because the critically-important PTIS values were not known,  $PTIS = 3$  (a modest degree of departure from both plug flow and complete mixing), was used. The resulting annual median k-values was 83 m/yr. There are not pronounced seasonal effects or temperature effects for fecal coliform removal in the FWS treatment wetland datasets that are currently available. Until more wetland data becomes available, it is recommended that global wetland fecal coliform reduction be regarded as independent of temperature and season.

## Summary of Estimated Wetland Water Quality Improvement

The presumed 3 hectares of lower wetland in the flow path of the effluent, if at a mean depth of 30 cm, would provide about two weeks detention in the wetland. Removal rate coefficients have been assumed to be at the low end of the probability distribution for nutrients, BOD and TSS; and these have been further reduced according to estimates of temperature effects caused by the 5°C water. **Table 1** shows the expected removals, and their relation to the data from other systems, and the implications for nutrient cycling. The forecast for water quality at the downstream end of the lower wetland are shown in **Table 2**.

**Table 2 Summary of Estimated Wetland Water Quality**

		From Lagoons	Wetland Outlet
TSS	mg/L	245	39
BOD	mg/L	230	38
TP	mg/L	5	4
Org-N	mg/L	20	9
NH <sub>4</sub> -N	mg/L	40	35
NO <sub>x</sub> -N	mg/L	0	11
TN	mg/L	60	54
TKN	mg/L	60	44
FC	#/100ml	10,000,000	157,000

## Wetland Cycles

Carbon, phosphorus and nitrogen are all cycled by the wetland ecosystem. It is prudent to examine whether the assigned rates lead to a reasonable set of ecosystem uptakes, releases and burials. These rely upon estimates of the amounts of biomass and nutrient contents, and so are not precise. The analyses are based upon estimates of what the ecosystem will be like. Such estimates cannot be precise, and consequently the analysis is "order-of-magnitude" only. The intent is to gain some insight into the relative importance of wetland processes that will likely be operating if the project is built.

### Carbon Cycling

The growth, death and decomposition processes are referred to as part of the wetland carbon cycle, but more than carbon is involved. However, most vegetation and other wetland organisms are about forty (40%) percent carbon, so either dry biomass or carbon serves to track the amount of material involved. Carbon itself is withdrawn from atmospheric sources, as carbon dioxide for photosynthesis. Likewise, carbon is returned to the atmosphere as methane from anaerobic mechanisms, or carbon dioxide from oxidative processes (respiration included). The important estimation quantities are:

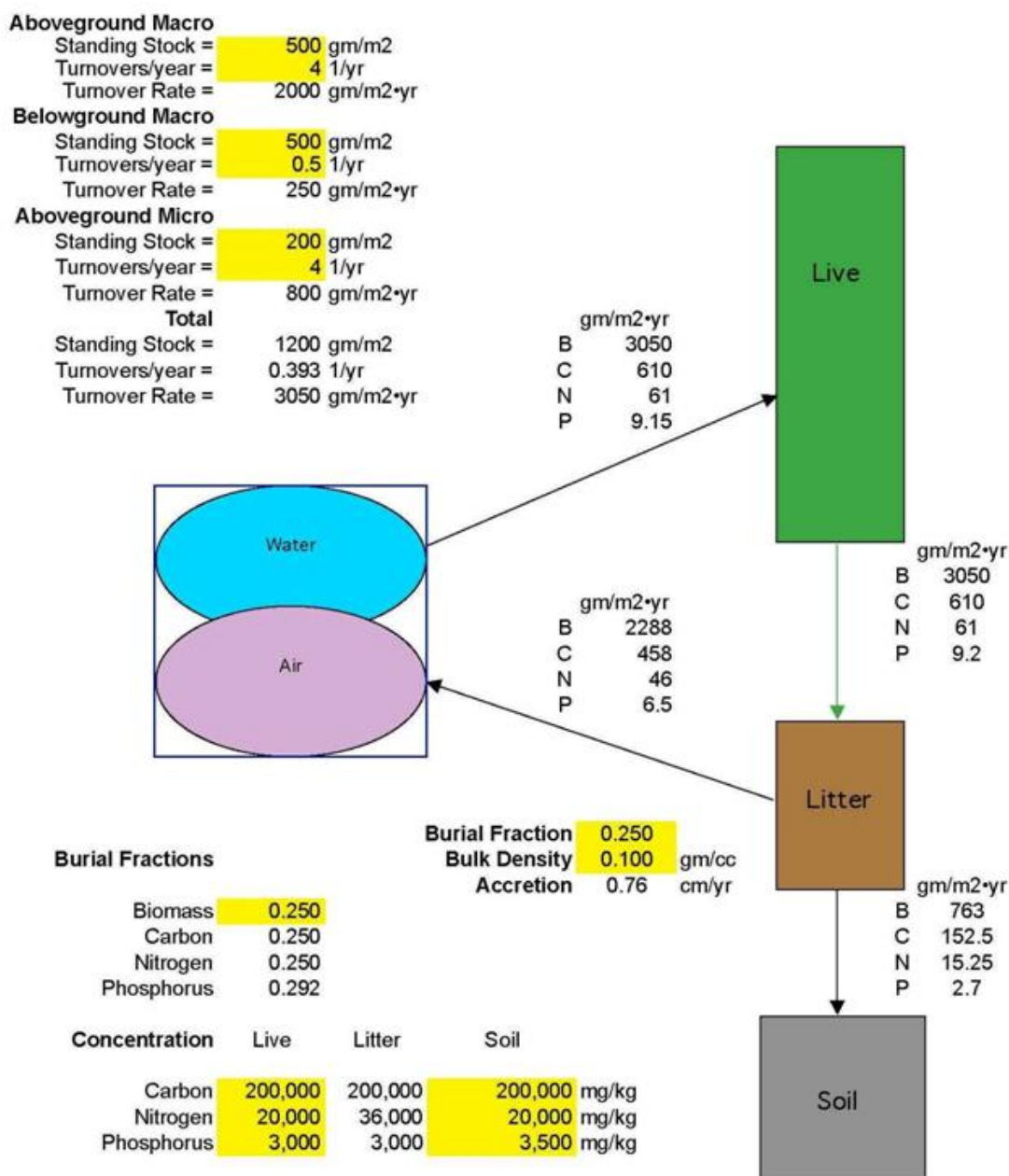
- Annual growth rate,  $\text{g/m}^2\cdot\text{yr}$  (standing crop phytomass = necromass + biomass times turnovers per year);
- Annual burial fraction (undecomposable residual fraction).

The assumption is made that nutrients taken up, but not buried as accretion, are returned to the water column of the wetland. For nitrogen, this is a maximum estimate, because microbial processes in above-water tissues can transfer nitrogen to the atmosphere without entering the water. These phytomass quantities, together with phytomass nutrient content (% or mg/kg) allow checks on the empirical removal calculations. The wetland carbon cycle is also critical to observed performance, as it relates to sediment oxygen demand and to the carbon supply for denitrification. The implied supply constraints of this carbon cycle are examined in the constraint check section.

The estimates for Kimmirut are given in **Figure 2**. The maximum standing stock of vegetation is assumed to be  $500 \text{ g/m}^2$  for each of above and below ground plant parts. Additionally, there is presumed to be  $200 \text{ g/m}^2$  of micro-scale biota, including bacteria, algae, and similar. The turnover of the above ground material is in the growing period of about three months, but roots are assumed to turn over more slowly, over a period of two years. Nutrient contents are assigned in accordance with a large body of observations, from warmer climates. About twenty-five (25%) percent of the biomass is assumed to be undecomposed. This carbon cycle is much smaller than would be observed in a warmer climate for similar water conditions, by about a factor of four or five.

### Phosphorus Cycling

For phosphorus, the calculated removal is represented as a large uptake, in major part balanced by the return of soluble P from tissue decomposition. For a stable ecosystem past startup, the net P removal associated with the k-rate calculation is assigned to accretion. Sorption and building of additional biomass no longer are sinks for phosphorus.



**Figure 2. Carbon Processing in Wetland**  
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**Figure 3** shows the estimate of the phosphorus cycle in the wetland. Because the nutrient concentrations are high (5 mg/L P and 60 mg/L TN), it is anticipated that the nutrient rich condition will prevail. The biogeochemical cycle removes  $33 \text{ gP/m}^2\cdot\text{yr}$  from the water column, a large fraction of the loading to the wetland of  $46 \text{ gP/m}^2\cdot\text{yr}$ . The k-rate calculations indicate that  $10 \text{ gP/m}^2\cdot\text{yr}$  are removed, and hence it is deduced that  $(33 - 10) = 23 \text{ gP/m}^2\cdot\text{yr}$  are returned to the water from decomposition and leaching of the biomass. The removal to accretion is thus  $10/33 = \text{thirty (30\%) percent}$  of the biomass uptake. These uptakes and returns involve above ground plant parts, below ground plant parts and microbes and algae.

### Nitrogen Cycling

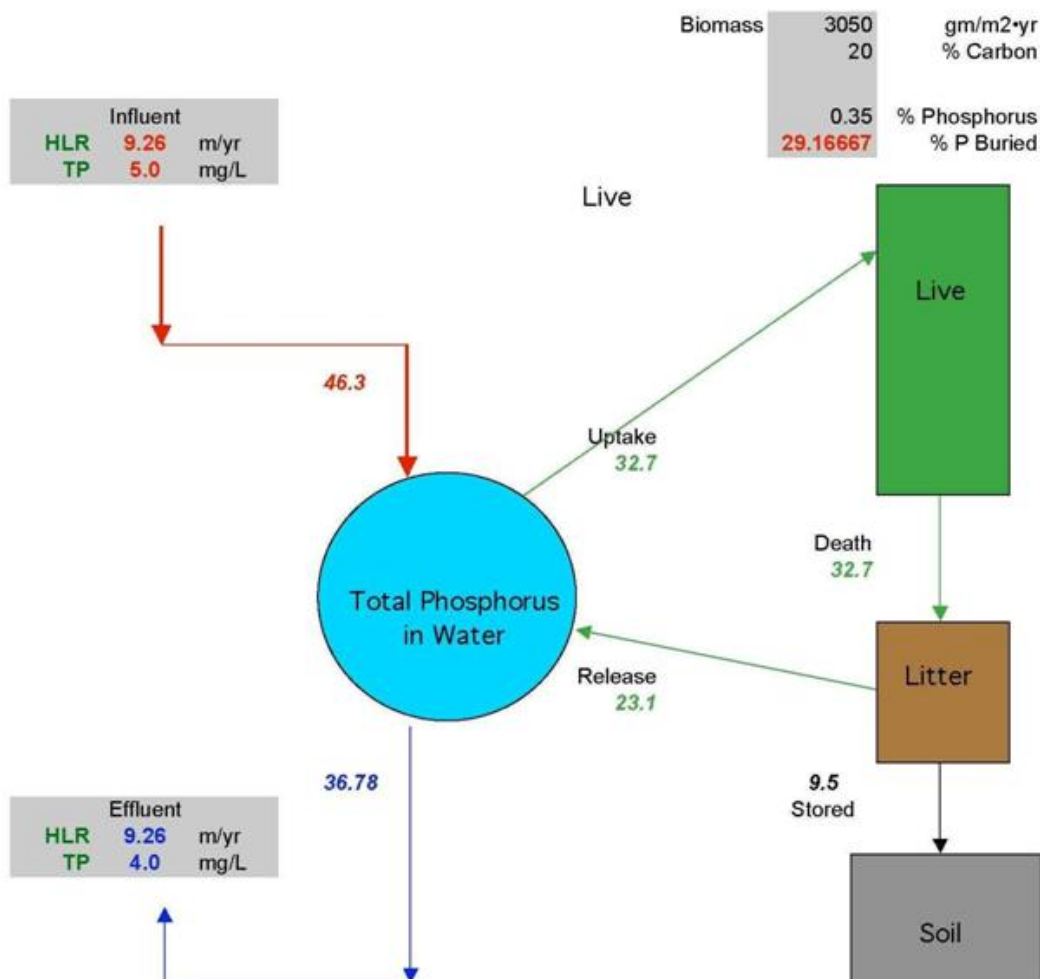
For nitrogen, a more complex situation occurs as a result of the multiple speciation of water column nitrogen. The water column contains organic, ammonium and nitrate nitrogen. Their interconversions are computed from the empirical k-rates. The biogeochemical cycle is linked in a manner analogous to phosphorus, but with abstraction from both the ammonium and nitrate pools in the water, and return from both the ammonium and organic pools in the water. This allocation recognizes a split of plant uptake between ammonium and nitrate, and the fact that decomposition processes produce organic nitrogen. The nitrogen content in accreting sediments is known from extensive data from treatment wetlands to range from ca. 1.0 to 2.5% dry weight. A higher value would be associated with nutrient-rich systems. Again, because the k-rate calculations are independent of the cycle calculation, there is one degree of freedom, which for nitrogen is taken to be an assumed percentage of the cycled nitrogen that is buried in accreting sediments. A nitrogen deficit may be assumed to be supplied by fixation, a process known to occur in N-deficient wetland environments.

As indicated in **Figure 4**, the incoming nitrogen is presumed to be mostly organic (20 mg/L) and ammonia (40 mg/L), comprising all of the 60 mg/L of TN. The nitrogen loading to the wetland is over  $555 \text{ gN/m}^2\cdot\text{yr}$ , which is a high loading that places the system in the category of a microbial system. It is therefore expected that the biogeochemical cycling of nitrogen will not play an important role in the overall reduction. For this proposed performance example, the cycling of biomass nitrogen is small. The required nitrogen to build the annually cycled biomass is  $61 \text{ gN/m}^2\cdot\text{yr}$ , which is eleven (11%) percent of the nitrogen loading (**Figure 4**).

The ultimate fate of removed nitrogen in the forecast is apportioned to accretion [fifteen (15%) percent] and denitrification [eighty-five (85%) percent].

### Supply Constraints

Traditional chemistry assumptions indicate a requirement for carbon to support heterotrophic denitrification, and an oxygen requirement to support nitrification. The oxygen requirements for the predicted performance are calculated. Removal of BOD ( $4.9 \text{ gO/m}^2\cdot\text{d}$ ) and nitrification ( $2.25 \text{ gO/m}^2\cdot\text{d}$ ) appear to exert a fairly high demand, which must be supplied by the cascade into the wetland and by atmospheric reaeration. The predicted denitrification requires an estimated  $90 \text{ gC/m}^2\cdot\text{yr}$ , which is easily supplied by the  $667 \text{ gC/m}^2\cdot\text{yr}$  of the BOD removed. The vegetation will produce an equivalent amount of carbon via decomposition, and thus the carbon supply is more than adequate to fuel denitrification.



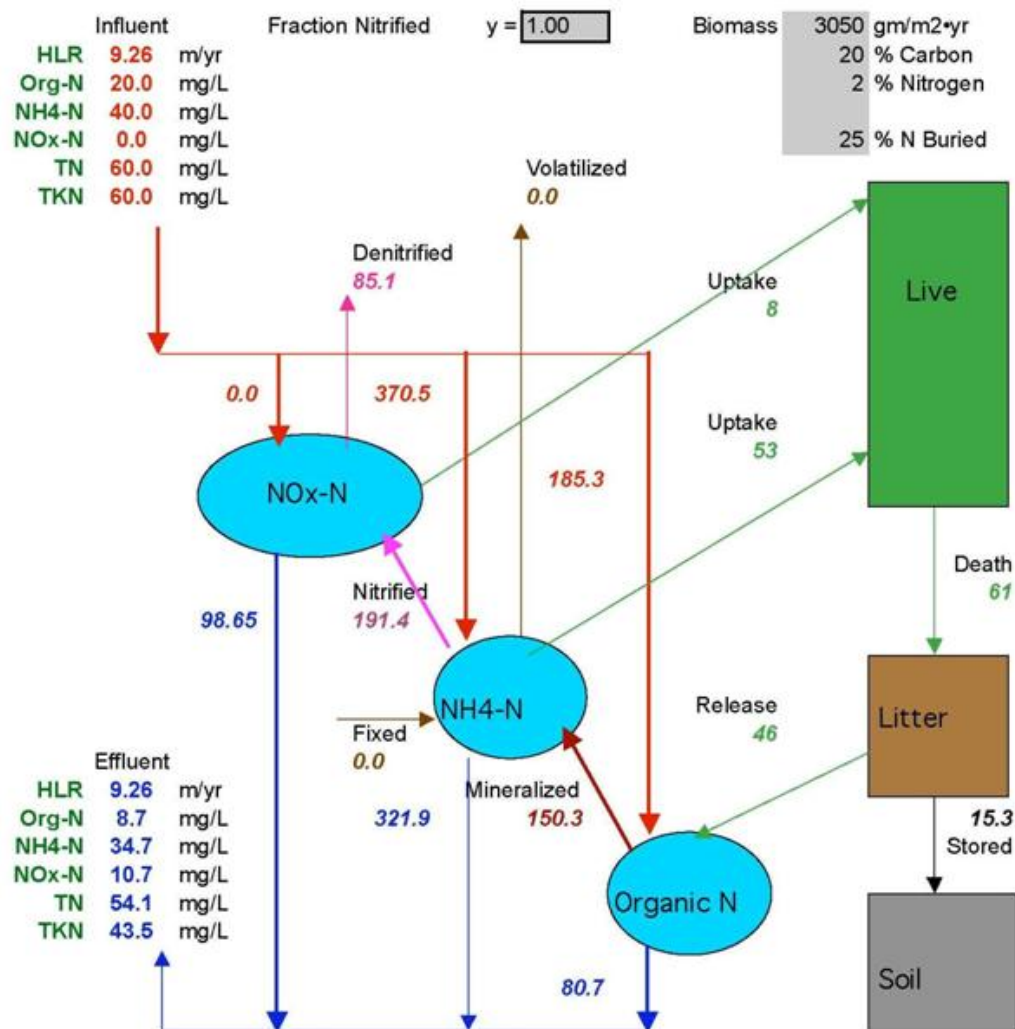
**Figure 3. Phosphorous Processing in Wetland**

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**Figure 4. Nitrogen Processing in Wetland**  
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## Inter-System Comparisons

Loading graphs allow inter-system comparisons for common constituents. These are a plot of the wetland outlet concentration (mg/L) versus the constituent loading to the wetland (kg/ha•d or g/m<sup>2</sup>•d). **Figures 5 – 8** show the position of the Kimmirut forecast with respect to the performance of a large number of operating treatment wetlands.

### *BOD and TSS*

The removal of these contaminants relies upon particulate settling and microbial activity by the general class of heterotrophic bacteria, as well as algal activity. Settling and trapping are known to be effective during all seasons and water temperatures. Bacteria are known to be effective in cold climates, and therefore there is no perceived penalty for the far northern site. The removal rate constants have been selected to produce average removals, and therefore the predicted performance is close to the central tendency of behavior of the comparison systems (**Figures 5 and 6**).

### *Nitrogen and Phosphorus*

The reduction of nitrogen in the wetland relies upon autotrophic nitrification to form oxidized nitrogen, and then upon heterotrophic denitrification. The nitrifying microbes are known to exhibit a rather slow build-up of populations, and are very sensitive to cold temperatures. Denitrifiers are usually present in larger numbers, and do not have a long period for growth, but they are also very temperature sensitive. Therefore, relatively low rate coefficients have been selected for nitrogen processing. The result is that the Kimmirut wetland shows lower TN removals than the comparison database, which reflects more southerly, warmer conditions (**Figure 7**).

Phosphorus removal does not involve microbes as heavily as nitrogen reduction, and a good share of the removal is due to the burial of a small fraction of the plant uptake. Therefore, P removal is tuned to the size of the growth cycle, which is anticipated to be rather small for this far northern site. Accordingly, a low removal rate coefficient has been used in forecasting, which results in a performance prediction well above the majority of comparison system data (**Figure 8**).

## Discussion

The foregoing analysis has been based on a set of conservative assumptions. It is virtually certain that the lagoon discharges will not spread over the entire 15 ha wetland, and it has been assumed that only twenty (20%) percent of the wetland (3 ha) will experience wastewater flows. It is also certain that considerable water flows already exist in the wetland, which might cause dilution, depending upon a complicated set of factors concerning the internal hydraulics of the wetland. However, this analysis assumes no dilution, with separate flow paths for existing waters and for wastewater.

The water reaching the wetland will have been subjected to detention in the upper and lower lagoons, which will provide some degree of water quality improvement. Additionally, water cascades from lagoon 1 to lagoon 2, and cascades again from lagoon 2 down into the wetland. Each cascade involves a fall of many meters, and will provide oxygenation of the water, and thus promote treatment. There will be some dilution in the lagoons, from regional rainfall, snowmelt and runoff, which will lessen the wastewater strength and increase the volume to be treated. Accordingly, the water quality entering the wetland has been estimated to be about half-strength, compared to the raw wastewater from the community.

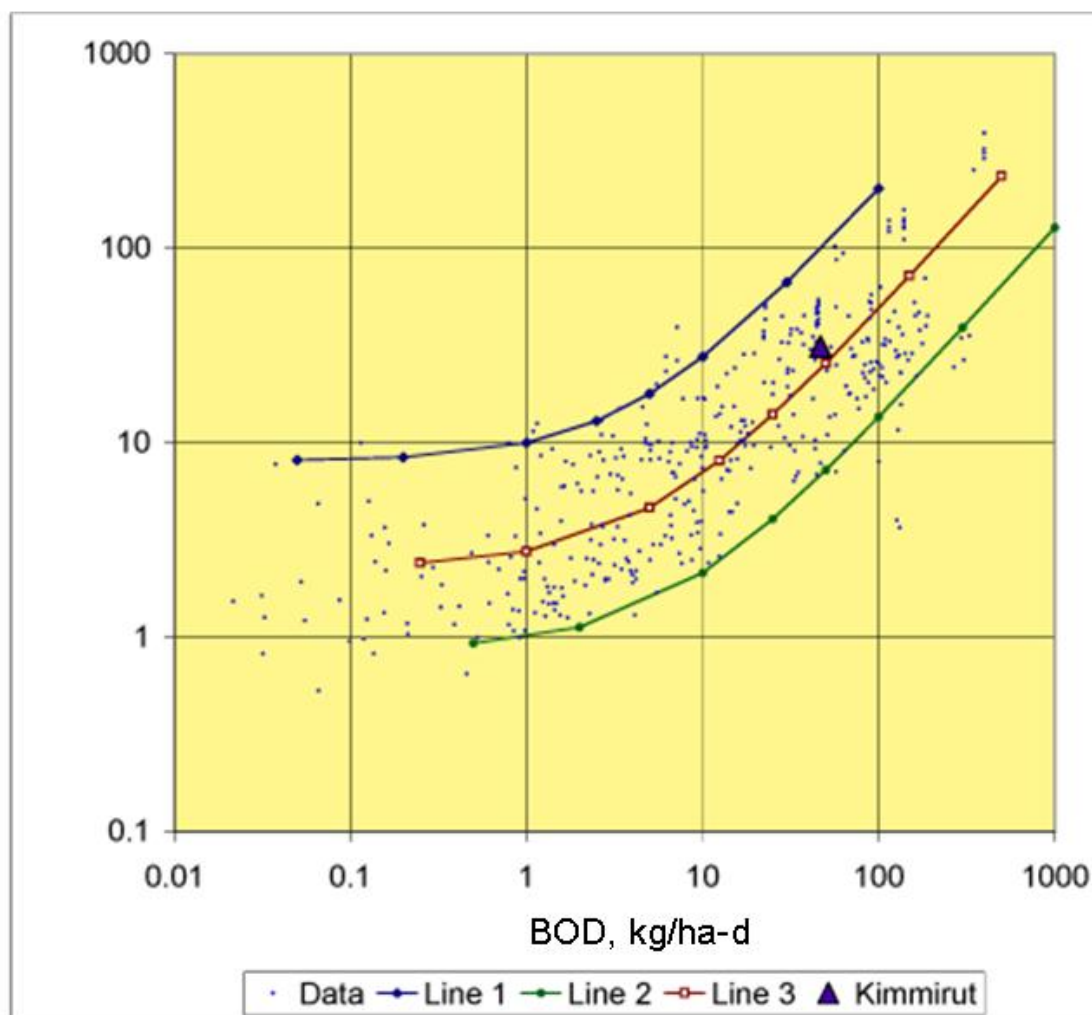


Figure 5. Loading chart for BOD in treatment wetlands. Each of the 383 points represents an annual average for one of 136 wetlands. The lines represent upper and lower bounds, and the central tendency. The Kimmirut forecast is close to the central tendency.

## Figure 5. Loading Chart for BOD in Treatment Wetlands Kimmirut Wetland Planning Study

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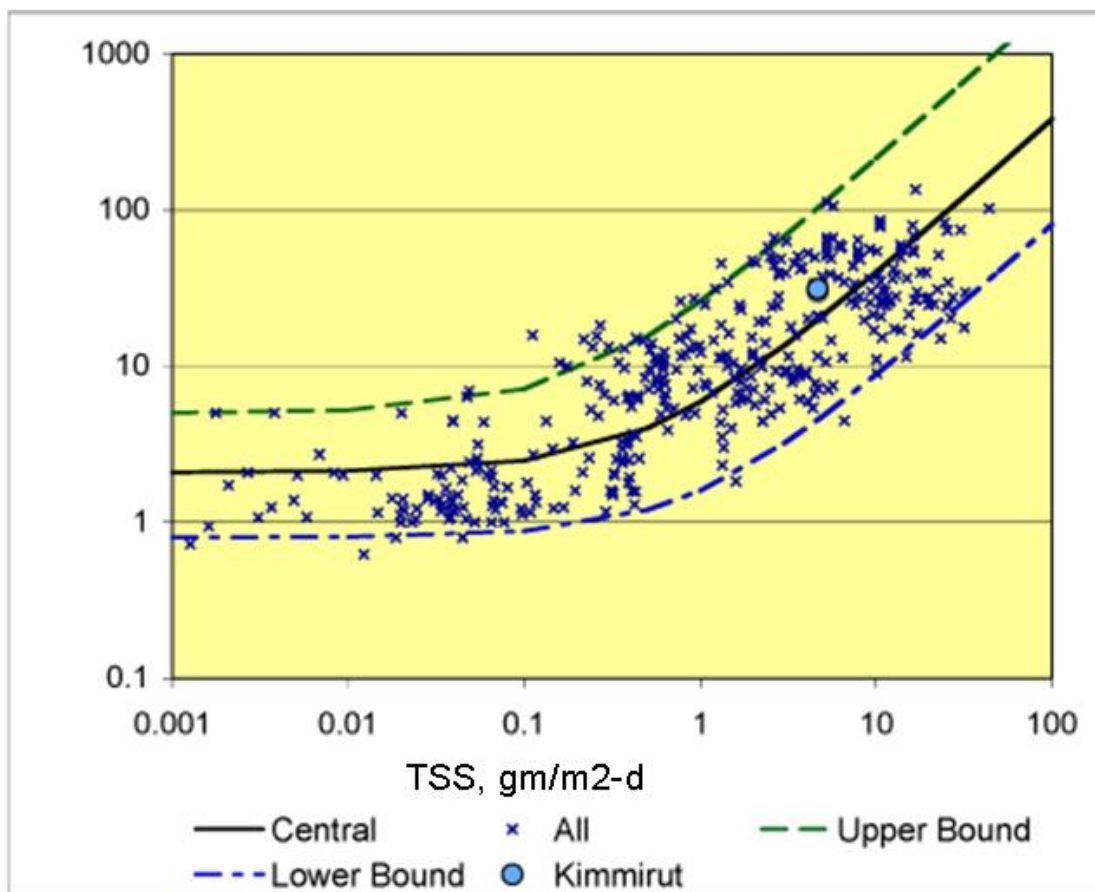


Figure 6. Loading chart for TSS in treatment wetlands. Each of the 449 points represents an annual average for one of 136 wetlands. The lines represent upper and lower bounds, and the central tendency. The Kimmirut forecast is close to the central tendency.

## Figure 6. Loading Chart for TSS in Treatment Wetlands Kimmirut Wetland Planning Study

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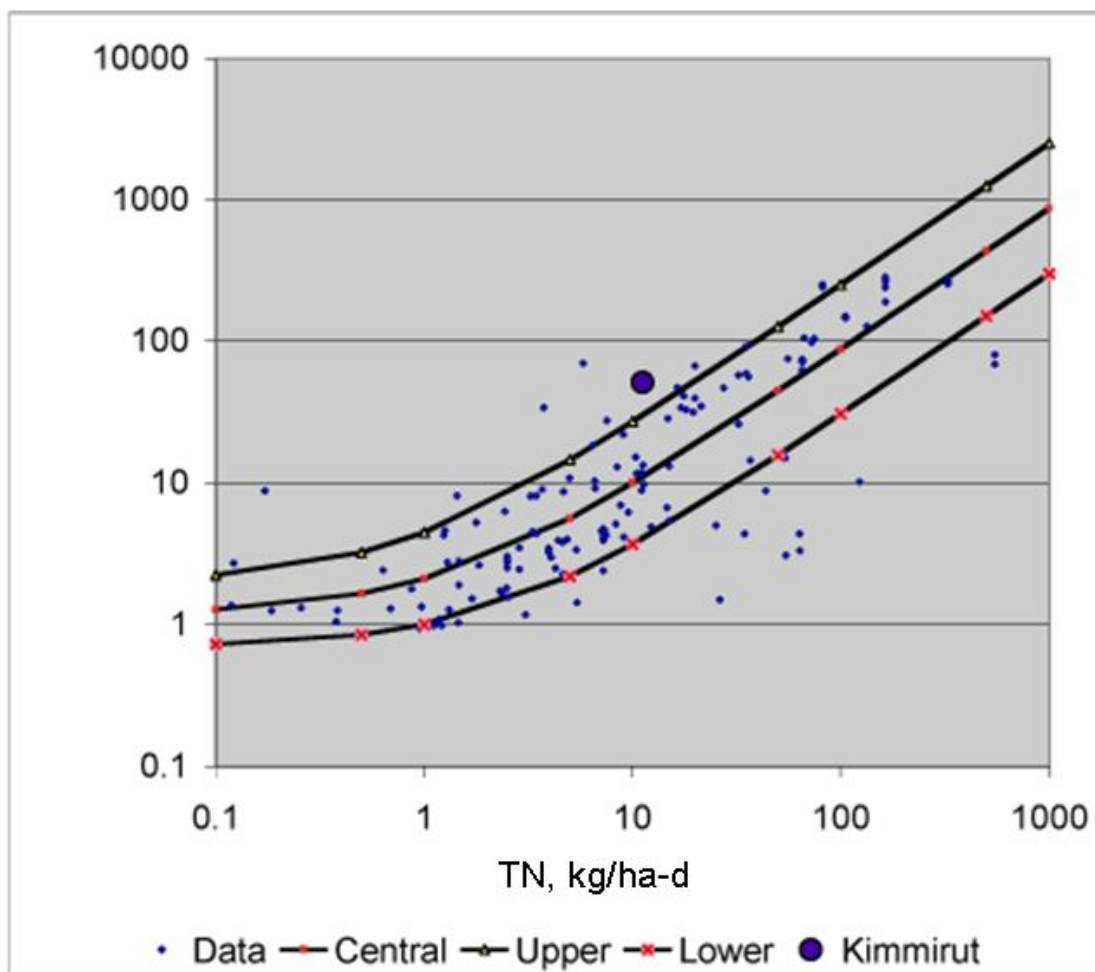


Figure 7. Loading chart for TN in treatment wetlands. Each of the 147 points represents a period of record average. The lines represent upper and lower bounds, and the central tendency. The Kimmirut forecast is well above to the central tendency.

## Figure 7. Loading Chart for TN in Treatment Wetlands Kimmirut Wetland Planning Study

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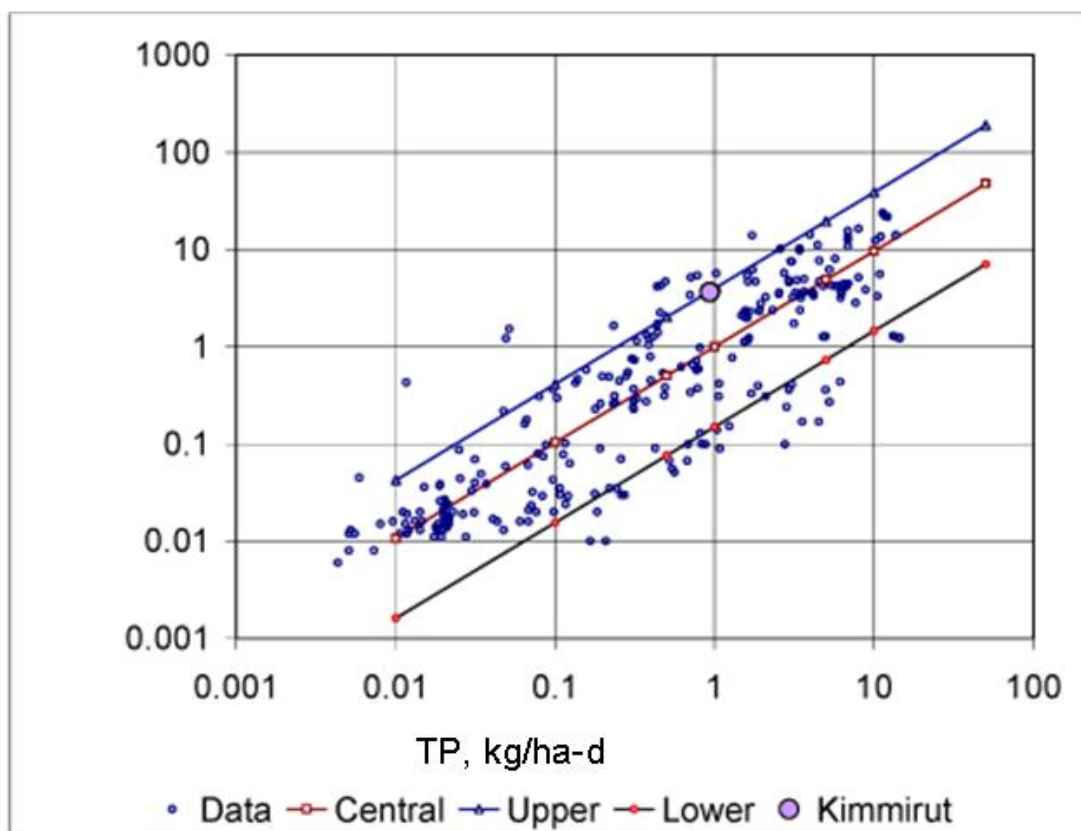


Figure 8. Loading chart for TP in treatment wetlands. Each of the 283 points represents a period of record average. The lines represent upper and lower bounds, and the central tendency. The Kimmirut forecast is above to the central tendency.

## Figure 8. Loading Chart for TP in Treatment Wetlands Kimmirut Wetland Planning Study

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Water quality improvement in the wetland has been calculated based upon a large amount of information from more southerly climates. However, the rates of biological processes have been assumed to be at the very low end of other experiences, including plant growth and nitrifier activity. Temperature coefficients have been used to reduce the rates of these processes, as indicated by the anticipated 5°C water temperature. The result of these assumptions is that BOD and TSS removal are likely to be comparable to wetlands in other climatic regions, but nutrient removal will be less. Some disinfection, or removal of pathogenic organisms is anticipated. There will be ample sunlight to promote UV disinfection in the wetland, as well as die-off due to cold temperatures. A two-log reduction [ninety-nine (99%) percent] is expected.

In short, the wetland will complement the two proposed lagoons and cascades, and provide good water quality improvement.

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